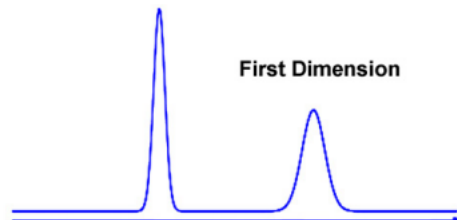


Multidimensional HPLC Tutorial

Part - 1

Introduction. Basic concepts – heart-cut vs comprehensive 2D-LC. Definition of orthogonality. The sampling problem of 2D LC.

What is Multidimensional HPLC?

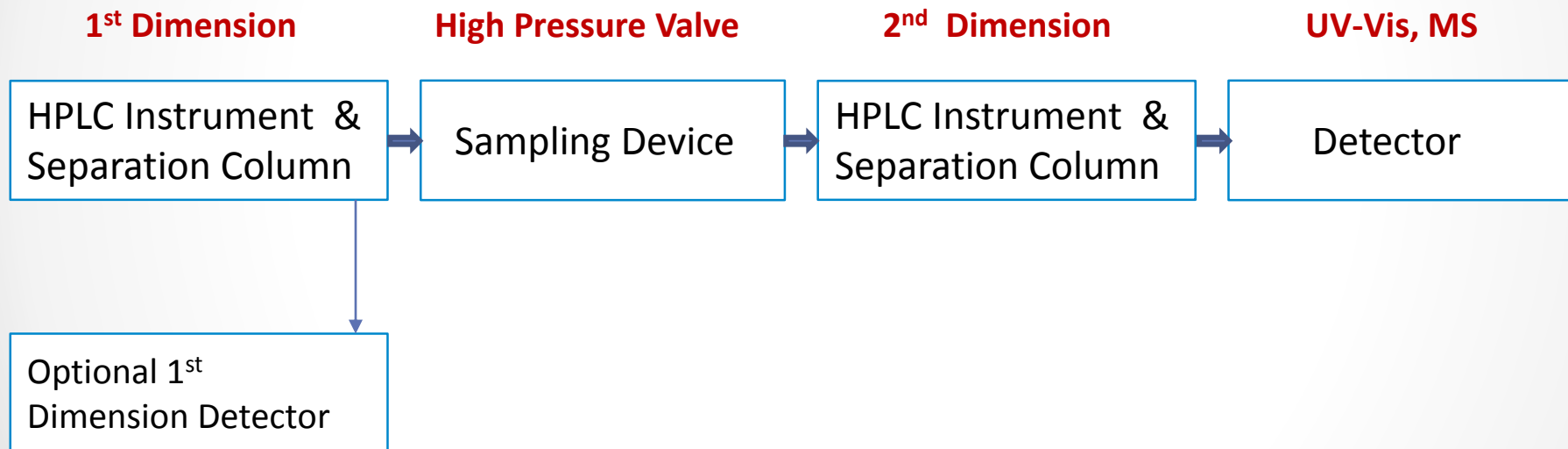


Peak capacity by the product of the number of bins

$$^1Z_p * ^2Z_p$$

What is Multidimensional HPLC?

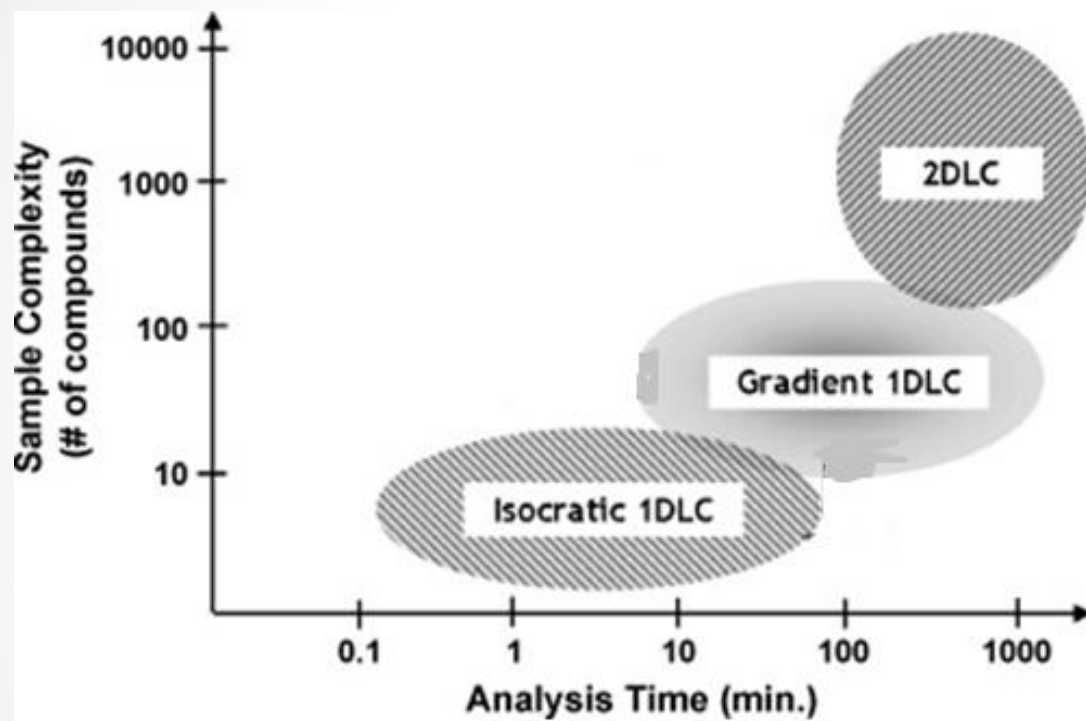
Simple Block Diagram



Applications Areas of MDLC

- Food, Beverages and Consumer Goods
 - Original Ingredients, Contamination, Proof Authenticity
- Proteomics, Metabolomics
 - Life Science Research
 - Biomarker discovery
 - Biopharmaceutical (originator or biosimilars)
- Environmental Analysis
 - Identification of Pollutants, Contaminants, Accidents
 - Polymers, Oligomers, Branching, Functional Group Analysis
- Forensics & Toxicology
 - Poison, Doping,
- Pharmaceutical Analysis
 - DMPK, metabolite identification
 - Traditional Chinese Medicine

Positioning of HPLC Techniques^{1,2}



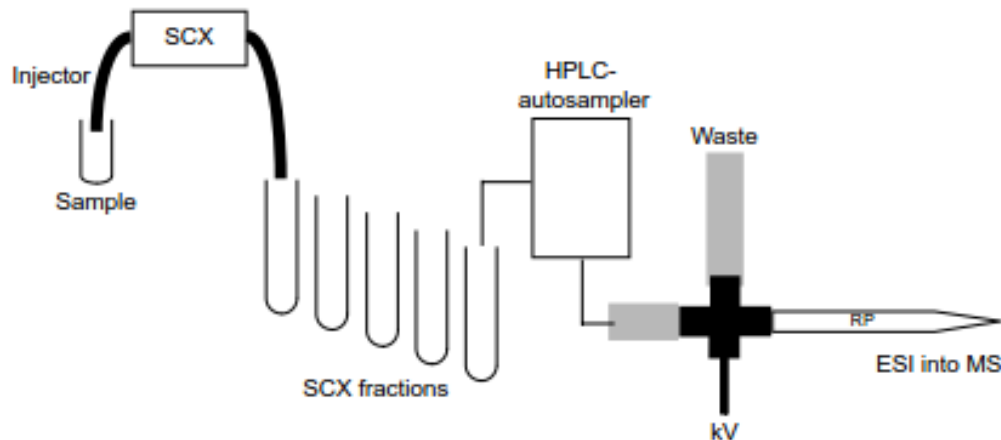
Adapted from ¹Stoll, D., University of Minnesota Ph.D. Dissertation, 2007, ²Stoll, D., et al., *J. of Chrom. A*, 1168, 3 (2007)

Important Nomenclature

- LC-LC heart-cut two-dimensional liquid chromatography
- LCxLC comprehensive two-dimensional liquid chromatography
- 1D 'first dimension'; for example, 1D column means 'first dimension' column
- 2D 'second dimension'; same 2D column means 'second dimension' column
- 1D denotes a one-dimensional system
- 2D denotes a two-dimensional system e.g. 2D LC
- 1t_r retention time for a given peak in the first dimension
- 1t_0 "dead" time of the first dimension conditions
- 1k retention factor of a given compound eluting from the first dimension column
- 1N the number of theoretical plates of the first dimension column
- 1w width of a peak eluting from the first dimension column
- 1R_s resolution of a peak pair eluting from the first dimension column
- 1Z_p peak capacity of the first dimension column
- n_c number of components in the sample

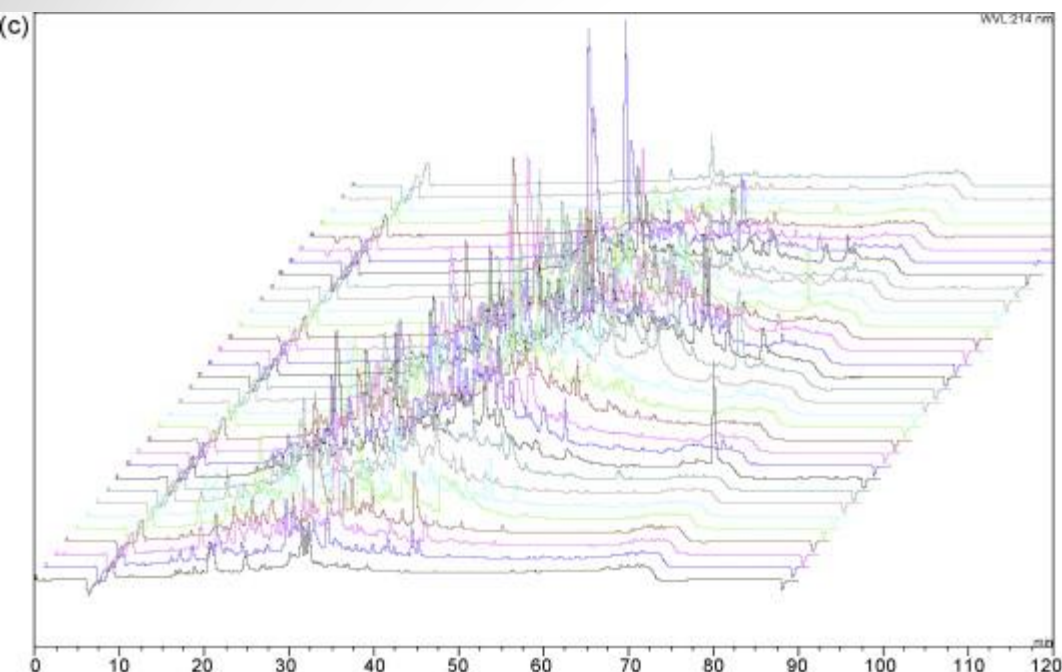
Principle Methods of 2D LC

- “Offline” methods (sequential)
 - Collect fractions from the 1st dimension separation, stored and re-injected in the 2nd dimension separation column in separate next run.



Picture taken from S.K. Swanson and M.P. Washburn, Drug Discovery Today, 10, 719 (2005)

Principle Methods of 2D LC



1st dimension:

150 mm L x 2.1 mm ID x 3.5 μ m XBridge phenyl column

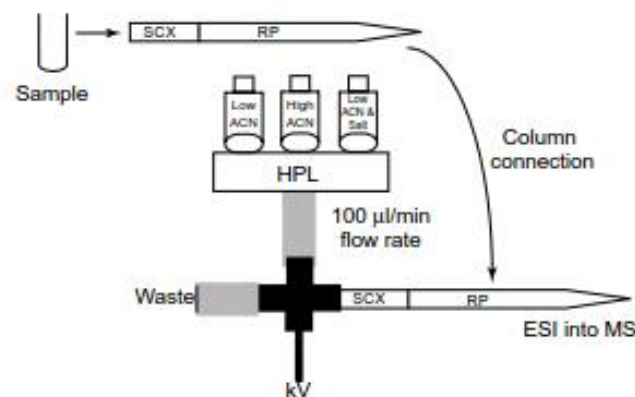
Offline fraction collection and reinjection in the 2nd dimension:

150 x 0.075 mm, 3 μ m Pepmap 100Å C18 particles

Total time required 40x2hrs!!

Principle Methods of 2D LC

- “Offline” methods (sequential)
 - Collect fractions from the 1st dimension separation, stored and re-injected in the 2nd dimension separation column later.
- “Stop-and-Go” methods e.g. MuDPIT*
(Multi-Dimensional Protein Identification Technology)
 - One column packed with a segment of ion exchanger and a larger segment of RP-phase. A pulsed salt gradient in IEX displaces a fraction of the sample onto the RP-column



Picture taken from S.K. Swanson and M.P. Washburn, Drug Discovery Today, 10, 719 (2005)

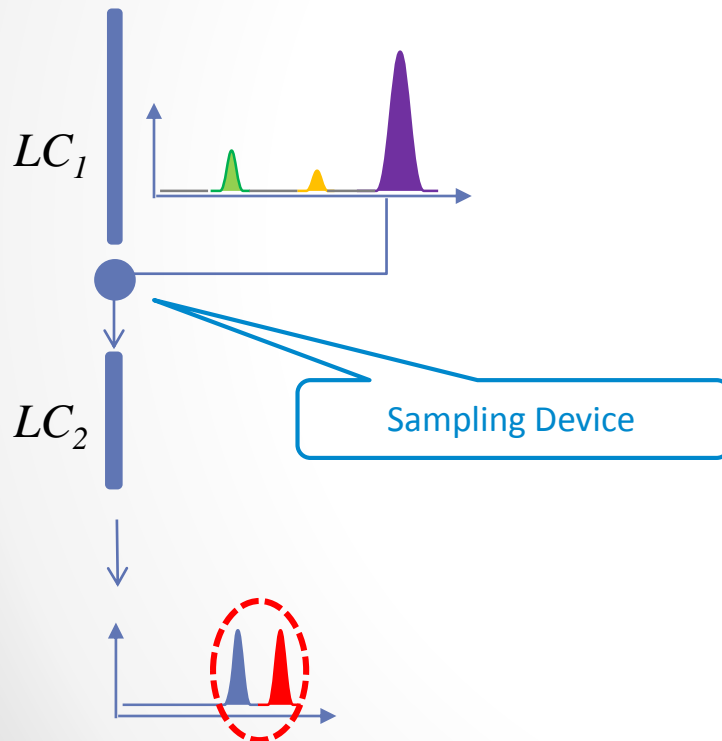
*J.R. Yates III et al., Int. J. of Mass Spectrometry 219 (2002) 245

Principle Methods of 2D LC

- “Offline” methods (sequential)
 - Collect fractions from the 1st dimension separation, stored and re-injected in the 2nd dimension separation column later.
- “Stop-and-Go” methods (e.g. Multi-Dimensional Protein Identification Technology)
 - One column packed with a segment of ion exchanger and a larger segment of RP-phase. A pulsed salt gradient in IEX displaces a fraction of the sample onto the RP-column
- “On-line” methods (parallel)
 - **Heart-cut:**
Selected fractions from the 1st dimension separation and intermediately stored on-line and delivered on-line to the 2nd dimension separation
 - **Comprehensive:**
Fractions are continuously taken from the eluate from the 1st dimension separation, intermediately stored on-line and delivered to the 2nd dimension separation

Principle Methods of 2D LC

Heart-cutting LC-LC

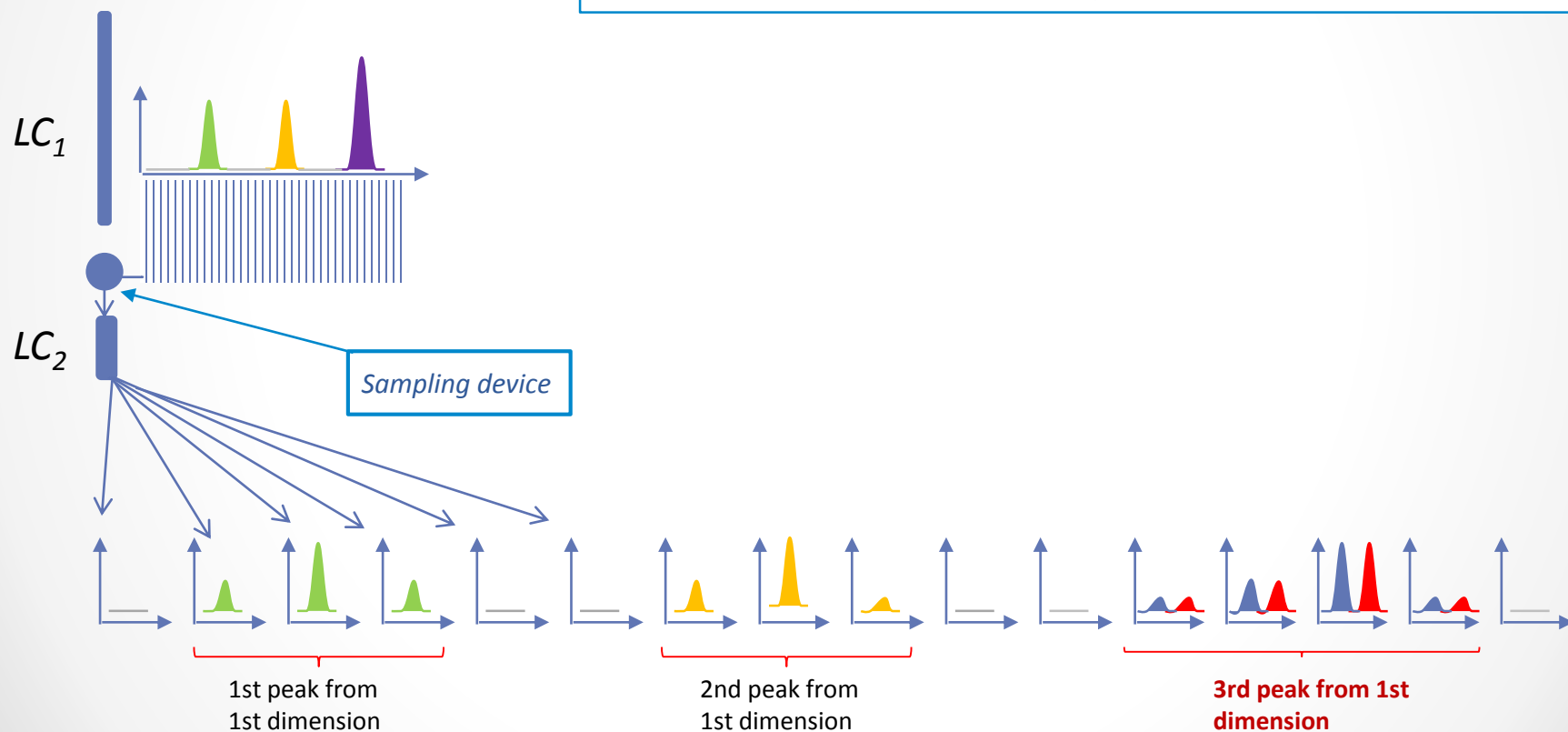


- Selected fractions of the 1st dimension separation are injected onto the 2nd dimension column
→ 1st dimension detector optional
- Long 1st dimension gradient separation possible
→ good data quality Limited information

Principle Methods of 2D LC

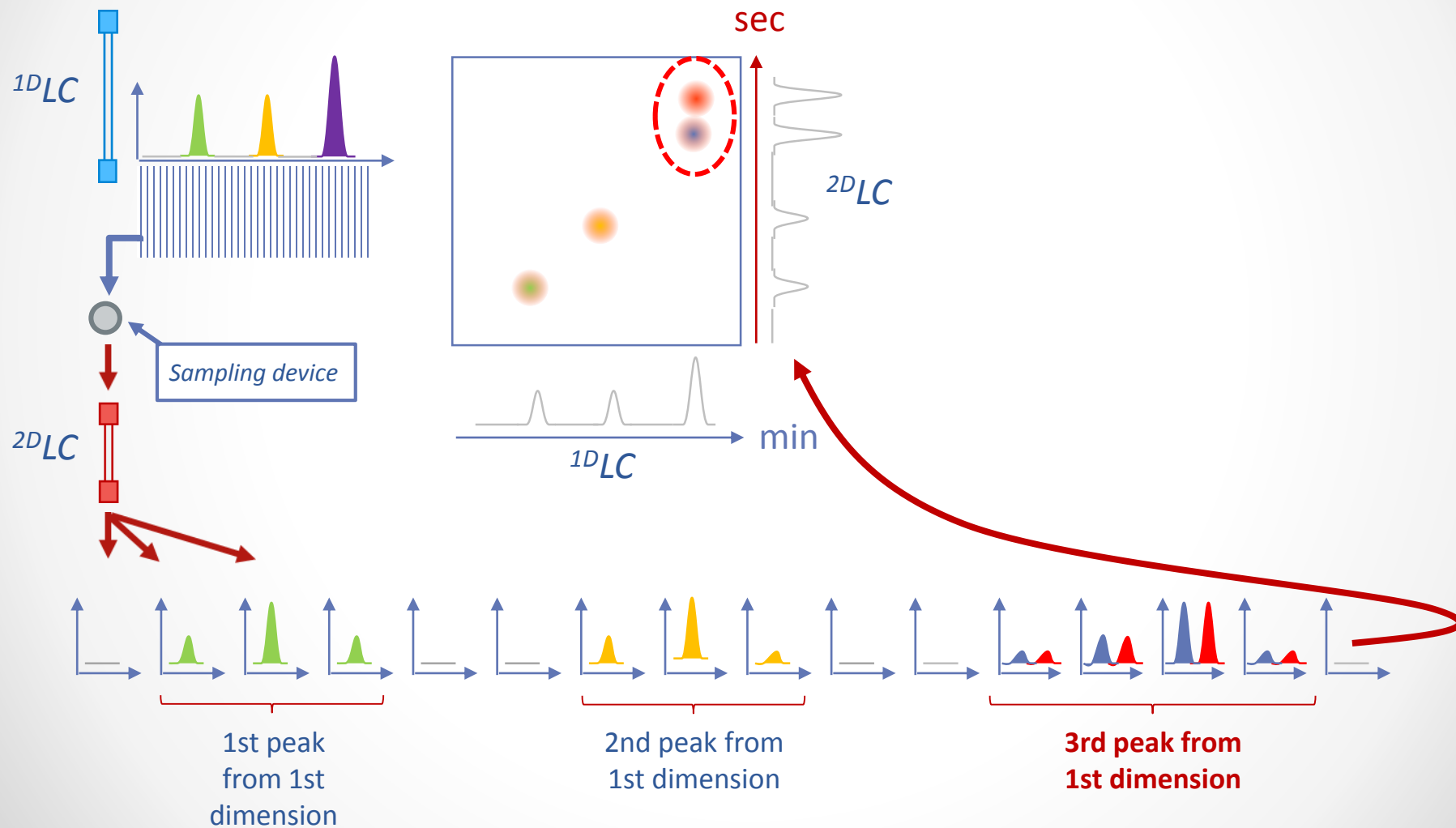
Comprehensive 2D-LC

- The whole ¹D effluent is continuously injected onto ²D column
- In the ²nd dimension (Ultra)Short 2D gradients are necessary mandating fast pumps & detector
- → Good data quality; full („comprehensive“) 2D information!



Principle Methods of 2D LC

Comprehensive 2D LC



Peak Capacity in 1D and 2D HPLC

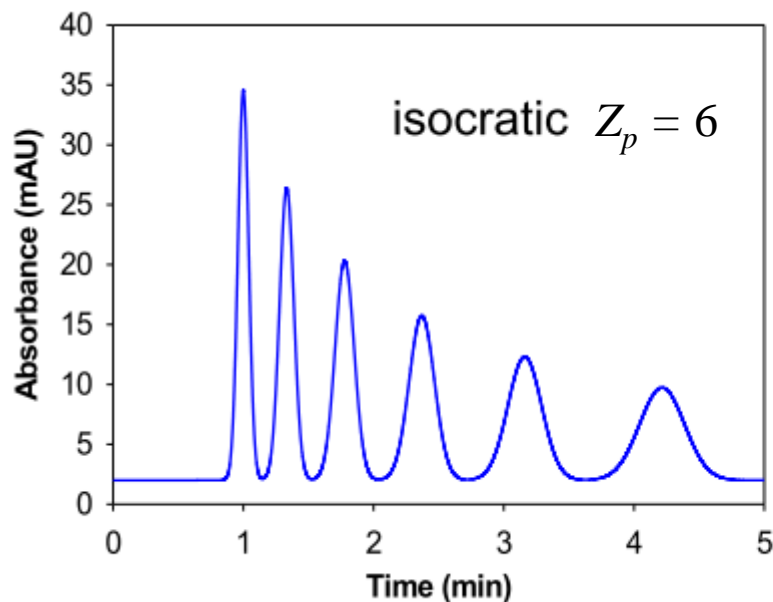
Why do Multidimensional HPLC?

- 1D HPLC does not give enough resolution to deal with complex samples ($n_c \gg 50$)
- Sample fingerprinting, classification, identification of contaminants, source of origin determination, detection
- Essential for “non-targeted” analysis (e.g. life science research*)
- Targeted analysis to isolate solutes from a complex matrix

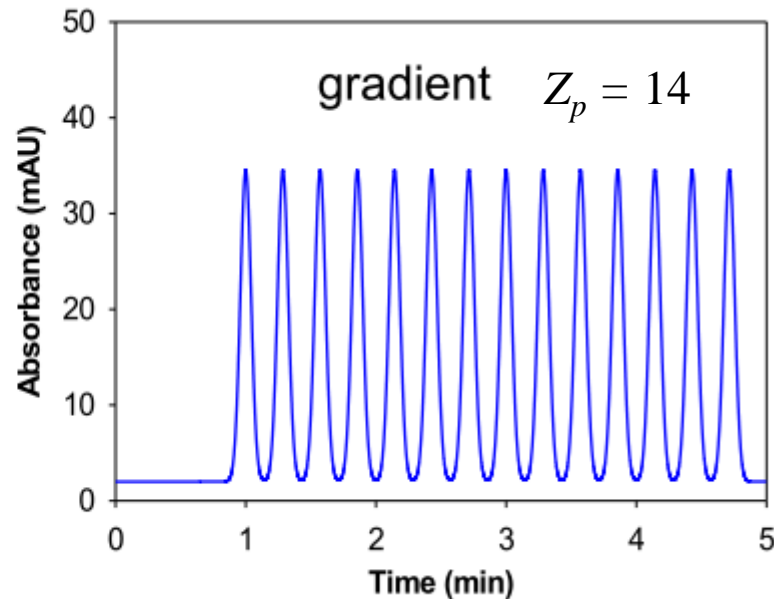
*Short anecdote

Peak Capacity (Z_p) in 1D HPLC

of peaks separated with equal resolution



Assuming linear solvent strength gradient*



$$Z_p = \frac{\sqrt{N}}{6R_s} \ln(1 + k'_{last}) + 1$$

Z_p : peak capacity
 k'_{last} : retention factor of the last peak
 R_s : required resolution (base line separation: $R_s \rightarrow 1.5$)

$$Z_p = 1 + \frac{t_g}{w_{av}}$$

LC column, $Z_p = 50$, $k = 10$, $N_{req} = \text{calculate}$

Peak Capacity (Z_p) in 1D HPLC

Limitations

- In real sample peaks are not regularly spaced but are randomly spaced.
- Peaks are not all the same size (height) thus the resolution criterion $R_s = 1$ may not show two maxima which only works for equal size peaks
- Davis–Giddings Statistical Model of Peak Overlap

$$s = n_c \exp\left(\frac{-2n_c}{Z_p}\right)$$

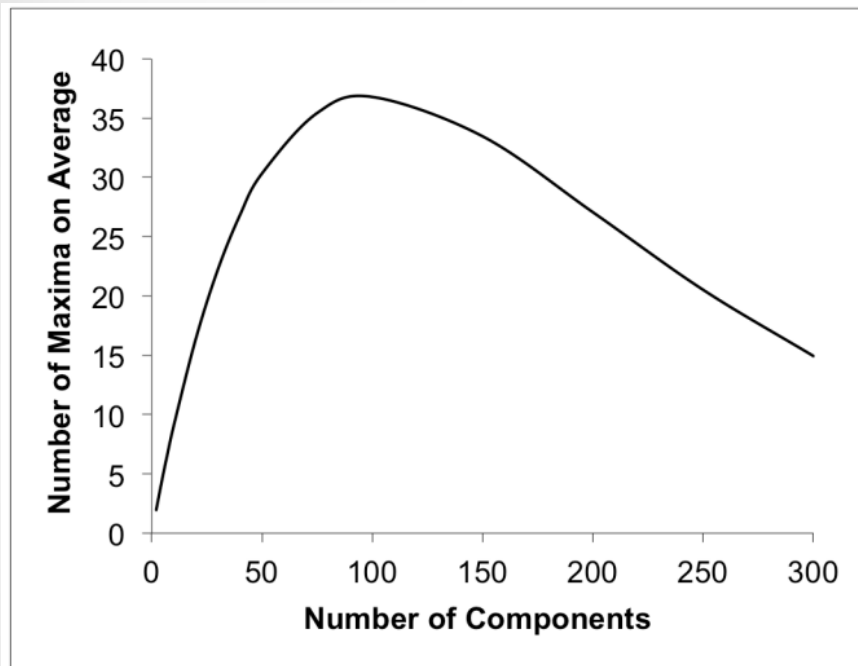
n_c = # of components in the sample

s = # of pure peaks in the chromatogram

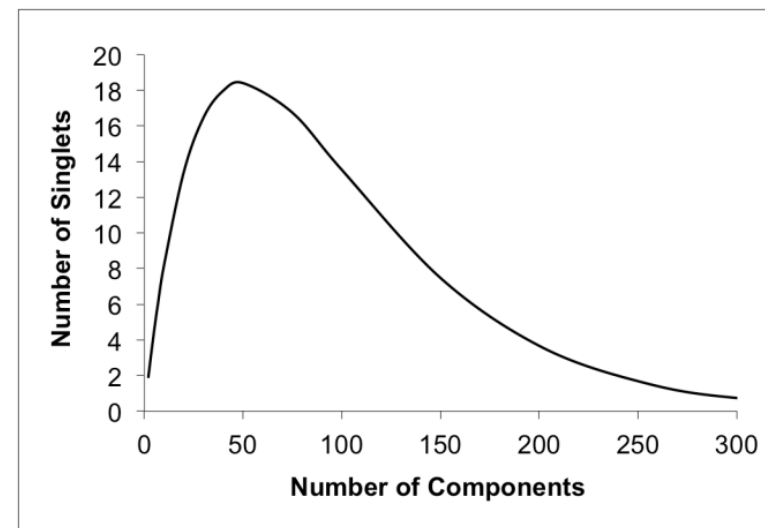
Peak Capacity (Z_p) in 1D HPLC

Davis–Giddings Statistical Model of Peak Overlap

of observed peaks ($Z_p = 100$)



of pure single peaks ($Z_p = 100$)

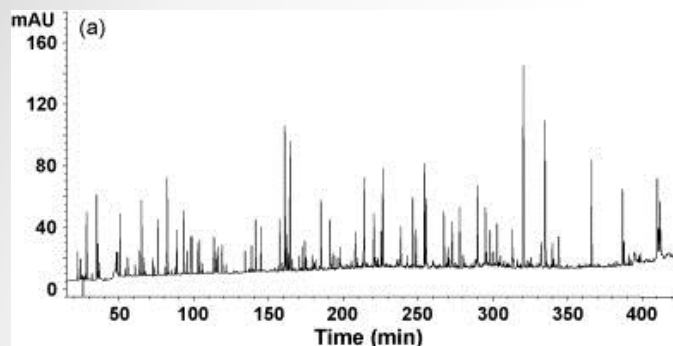


$$p_{max} = 0.37 Z_p$$

$$s_{max} = 0.18 Z_p$$

Peak Capacity (Z_p) in 1D HPLC

Practical Example

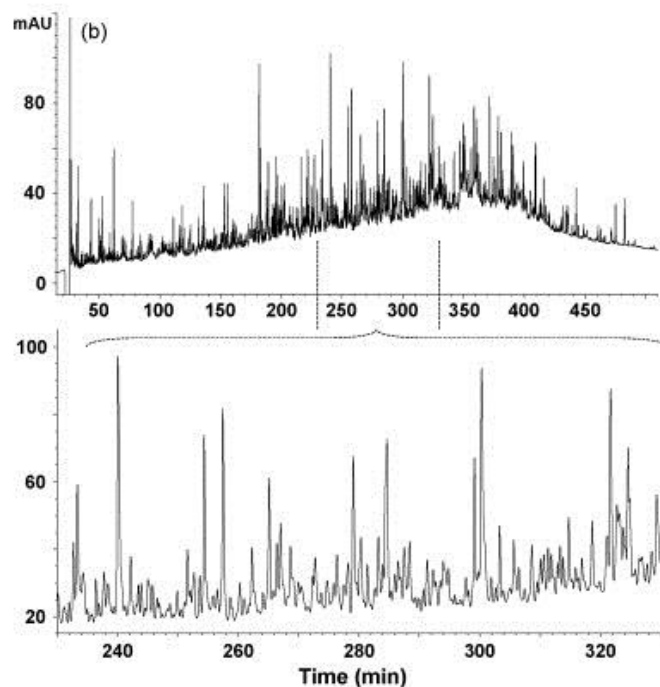


BSA (a) and a depleted human serum tryptic digest (b) on 8 250× 2.1 mm ID × 5 μ m Zorbax SB300-C18 columns.

Temperature 60 °C.

Mobile phase A 2% ACN, 0.1% TFA and mobile phase B 70% ACN, 0.1% TFA. Gradient slope 0.135% B/min, flow rate 200 μ L/min.

Detection wavl. 214 nm



Peak Capacity in Comprehensive 2DLC

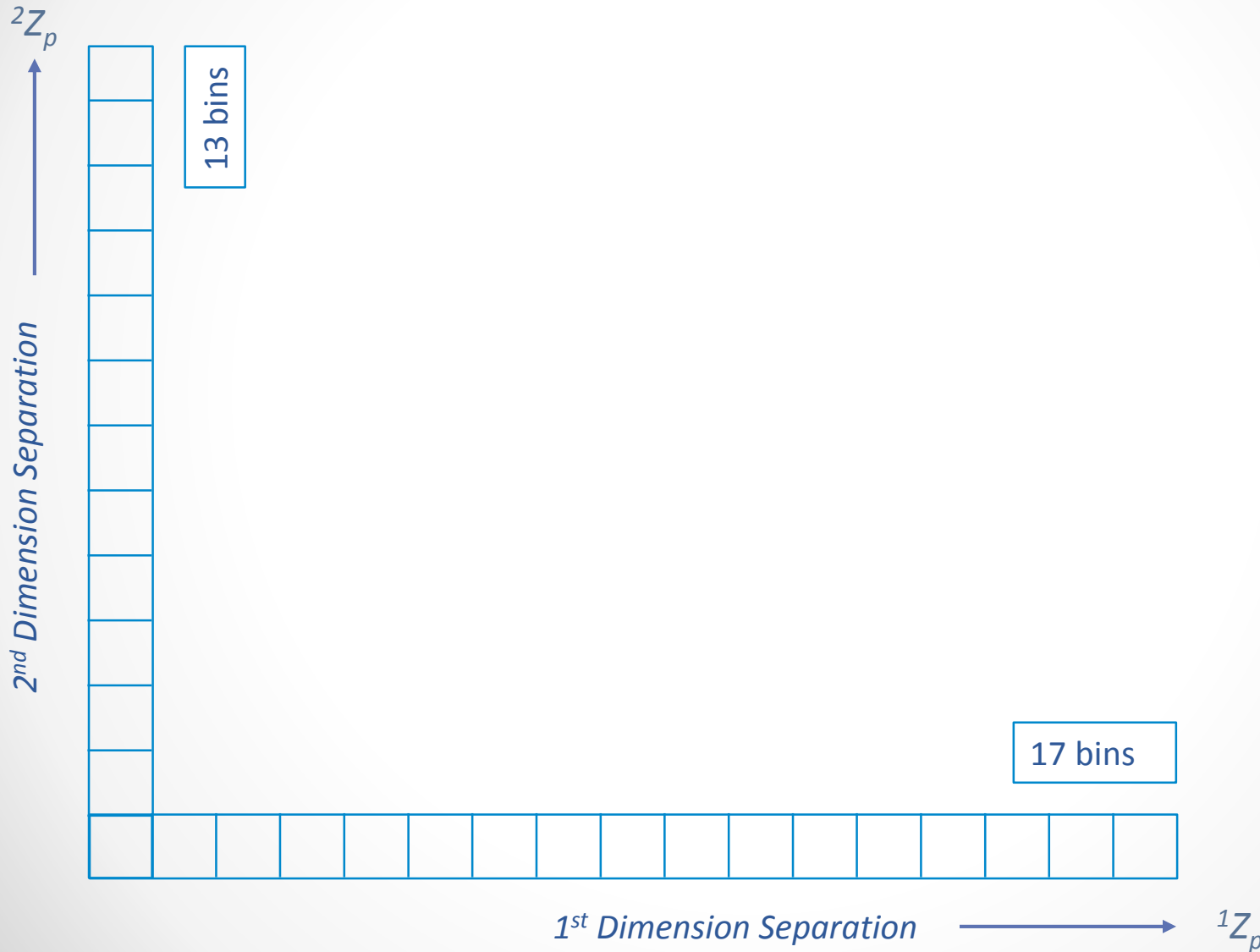
The geometric orthogonality concept



1st Dimension Separation →

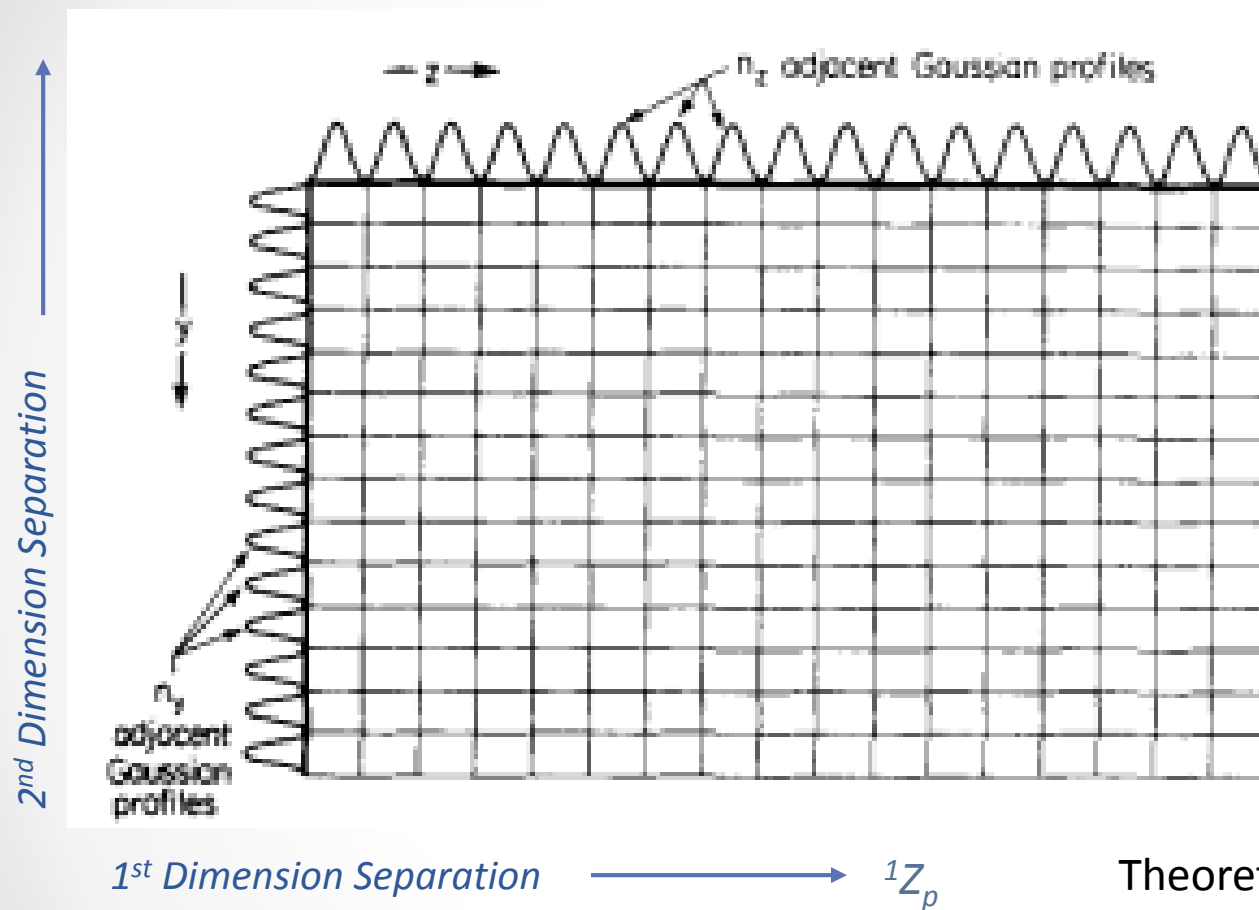
Peak Capacity in Comprehensive 2DLC

The geometric orthogonality concept



Peak Capacity in Comprehensive 2DLC

The geometric orthogonality concept



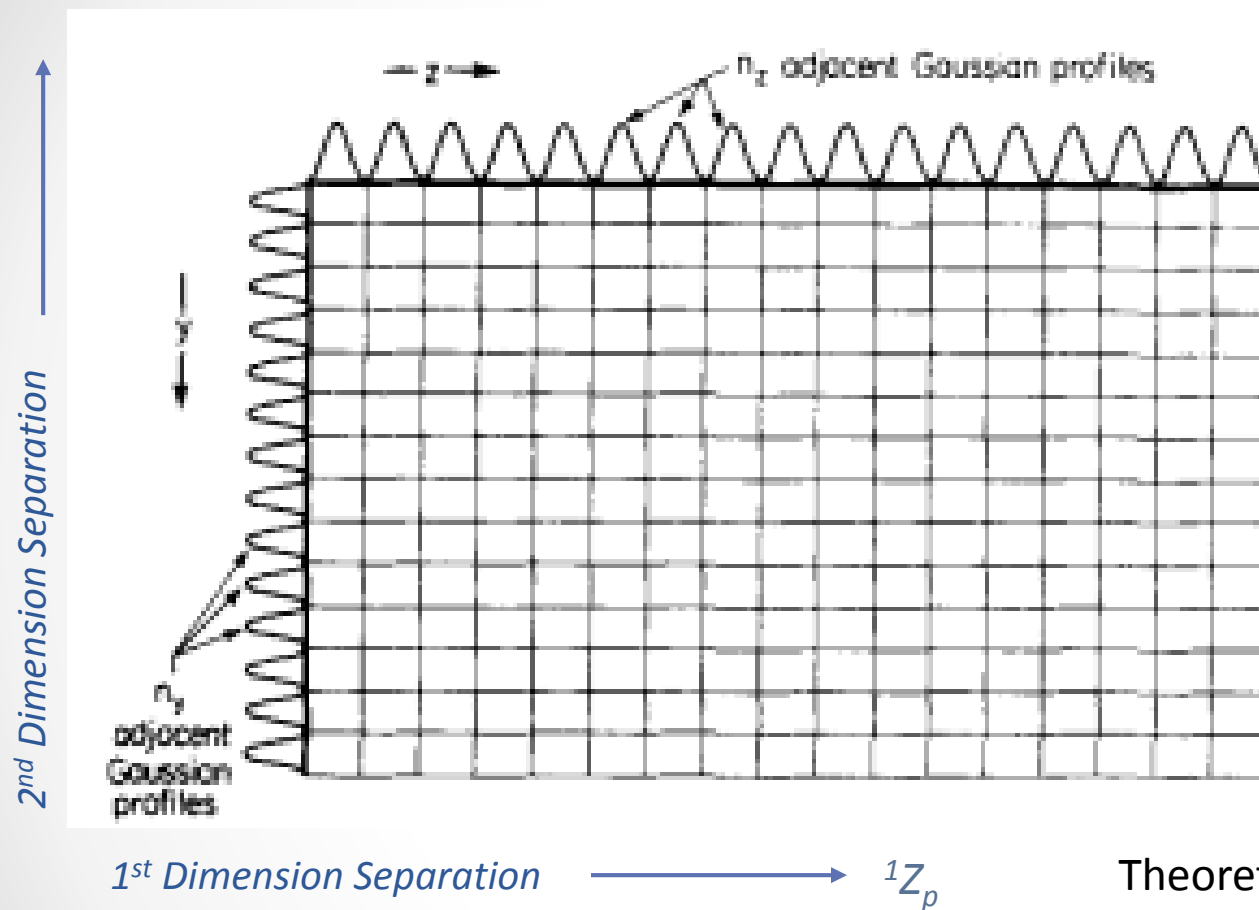
Theoretically:

$${}^{2D}Z_p = {}^1Z_p \times {}^2Z_p$$

The Giddings “Product Rule”

Peak Capacity in Comprehensive 2DLC

The geometric orthogonality concept



Theoretically:

$${}^{2D}Z_p = {}^1Z_p \times {}^2Z_p$$

The Giddings “Product Rule”

In this case ${}^{2D}Z_p = 17 \times 14 = 238!$
 For 1D separation, $N_{req} = \text{calculate}$

Peak Capacity in Comprehensive 2DLC

Giddings Criteria for the Product Rule

■ ORTHOGONALITY:

“First, the components of a mixture are subjected to two or more separation steps in which their displacements depend on different factors.”

- The retention of the sample solutes must be controlled by two (or more) different physical-chemical properties and the two separation systems must separate the species by different mechanisms. The retention of the component is describe by two or more retention times (by 2 dimensions)

■ SAMPLING:

“The second criterion is that when two components are substantially separated in any single step, they remain separated until the completion of the separation step.”

- Width of the sample (in time or volume units) from first separation
- Once the solutes are separated there must be no remixing (peak broadening) induced by doing the second separation

Requirements to Achieve Theoretical Peak Capacity in Comprehensive 2DLC

1. **ORTHOGONALITY** of separation mechanisms – This is a requirement imposed mostly on the **stationary phase** chemistry.
2. Peaks must cover **ENTIRE** separations “space”.
3. Separation gained in one dimension cannot be diminished by separation in the other dimension. Must sample **FAST!**

$${}^{2D}Z_p = {}^1Z_p \times {}^2Z_p \times \frac{1}{\langle \beta \rangle} \times f_{\text{coverage}}$$

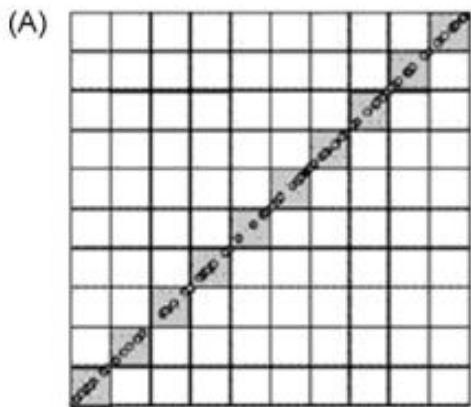
Under Sampling
Correction

Orthogonality

What is the most important factor?
How can we improve it?

Orthogonality in 2D LC

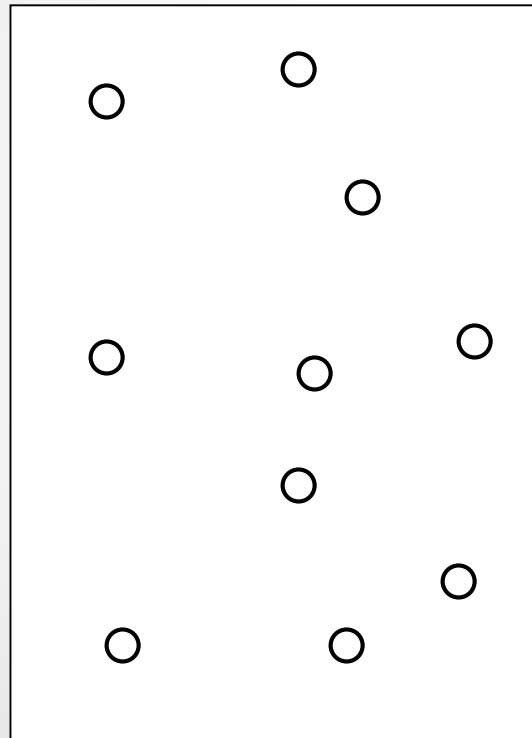
Orthogonality in Comprehensive 2DLC



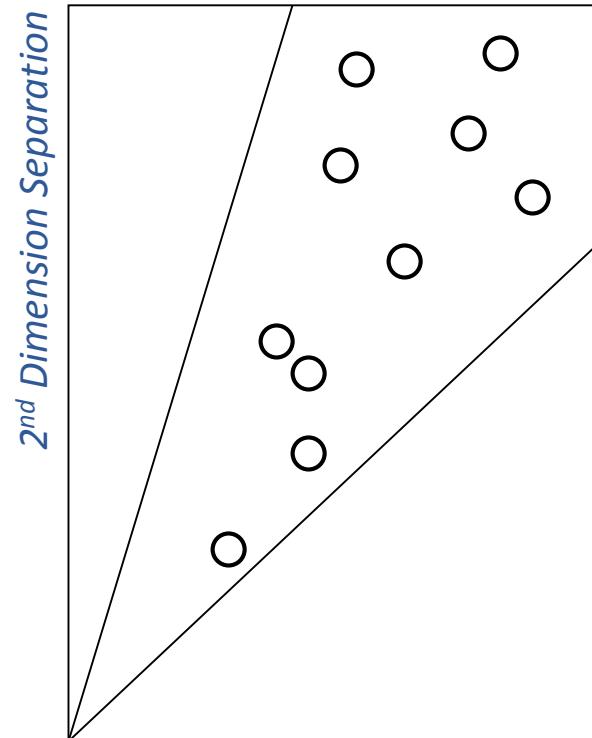
(A) Non-orthogonal system, 1D column is identical with 2D column. Area coverage represents 10% orthogonality.

Separation Space Utilization by Orthogonal and Correlated Mechanisms

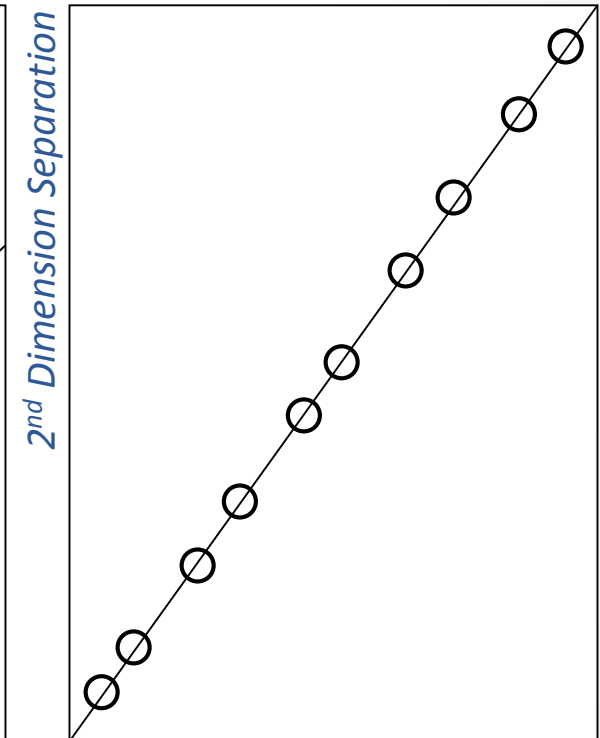
Orthogonal separations
uncorrelated separations



Orthogonality with partial
correlated separations



No orthogonality
separations correlated



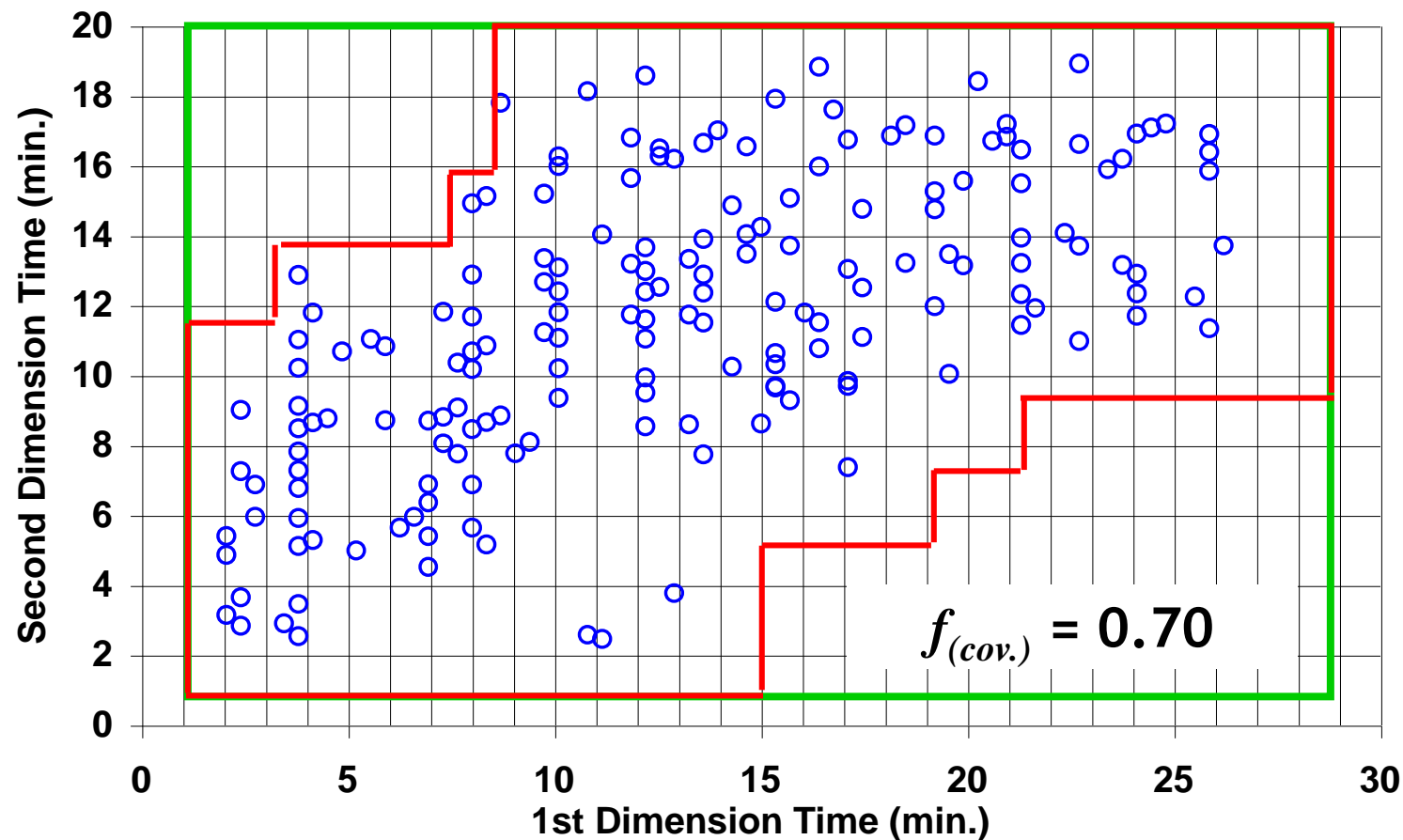
1st Dimension Separation

1st Dimension Separation

1st Dimension Separation

Fractional Utilization of 2D Space

(Stoll modified Gilar method)



Stoll, D.R., et al. *Anal. Chem.* 2008, 80, 268-278

Gilar, M. et al. *Anal. Chem.* 2005, 77, 6426-6434.

Problems with Coverage Estimation Methods

- Correlation coefficients don't measure available space.
- Stoll-Gilar type methods are subjective and depend strongly on grid size.
- Many other methods are complex and have critical parameters.

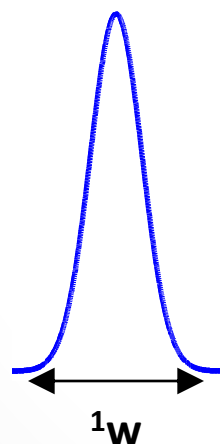
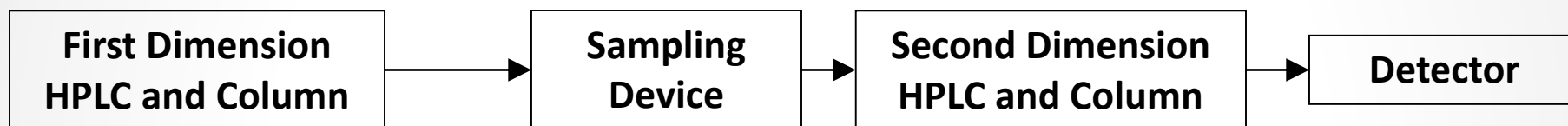
*For an overview, see the excellent comparison paper of Gilar and Schure, *et al.*

M. Gilar, J. Fridrich, M.R. Schure, A. Jaworski, Comparison of Orthogonality Estimation Methods for the Two-Dimensional Separations of Peptides, *Analytical Chemistry*. 84 (2012) 8722–8732.

The Sampling Problem in 2D LC

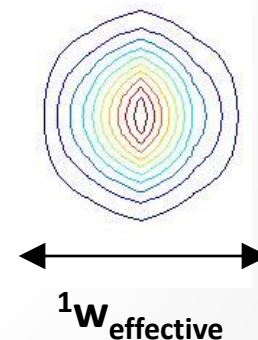
Giddings's Second Criterion

The width of a peak observed in a 2D chromatogram in the direction of the first dimension axis **after sampling** is effectively broader than the width of the peak that elutes from the first dimension column **before sampling**.



Time →

Second
Dimension Time ↑

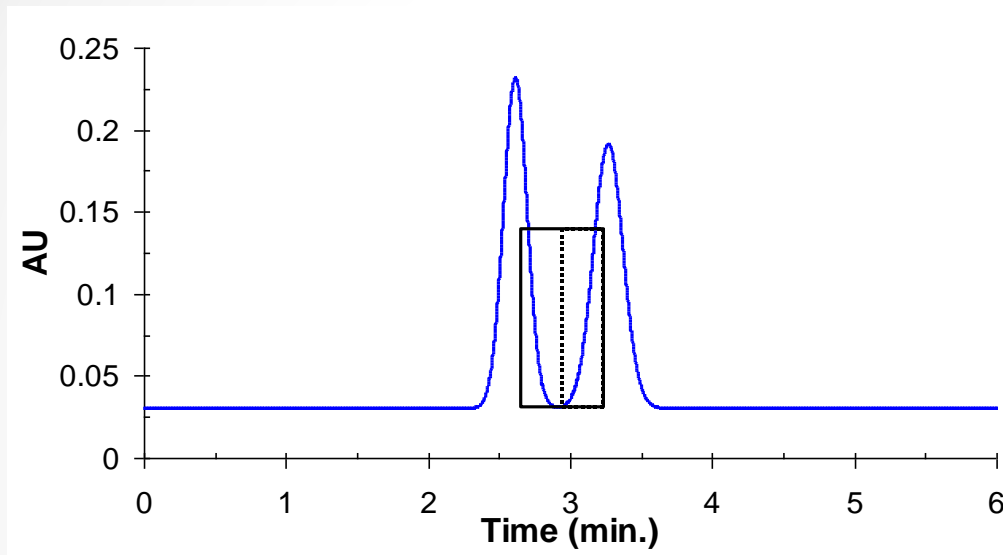


First Dimension
Time →

The Undersampling Problem*

The Murphy-Schure-Foley Criterion

*Slide courtesy of Prof. Pete. Carr & Dr. Dwight Stoll

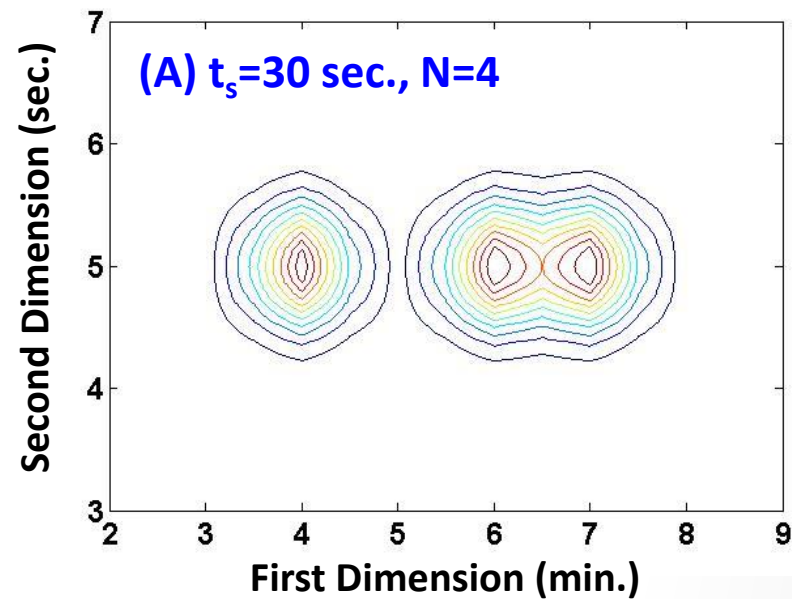
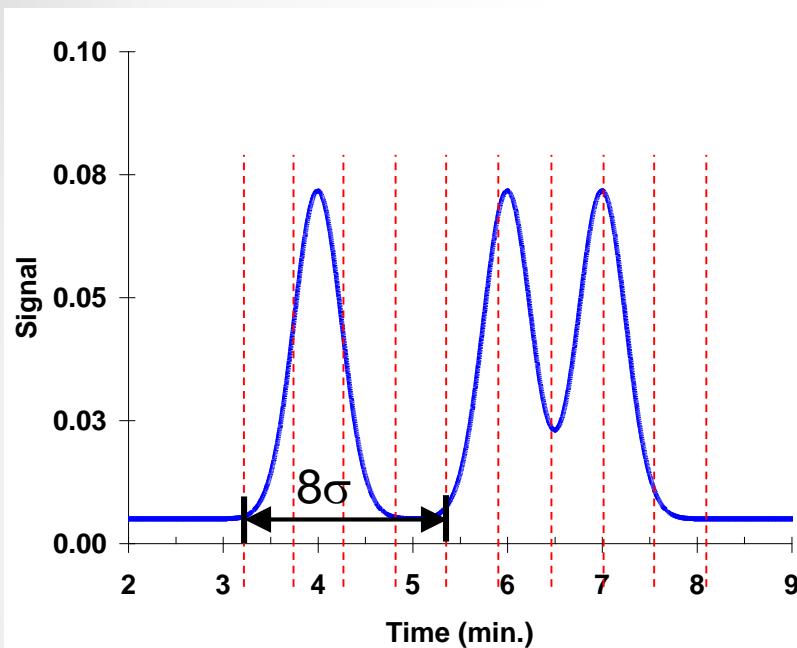


Clearly if we take a sample as indicated and inject it into a second dimension we will partially “un do “ the separation already accomplished in the first dimension.

According to M-S-F **one needs to take at least 4 samples** across the 8σ base width of each first dimension peak **to minimize the effect of undersampling**.

The Undersampling Problem

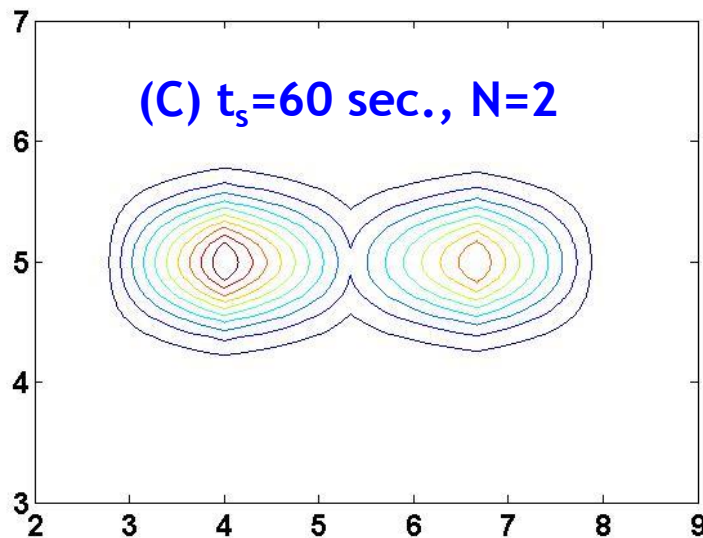
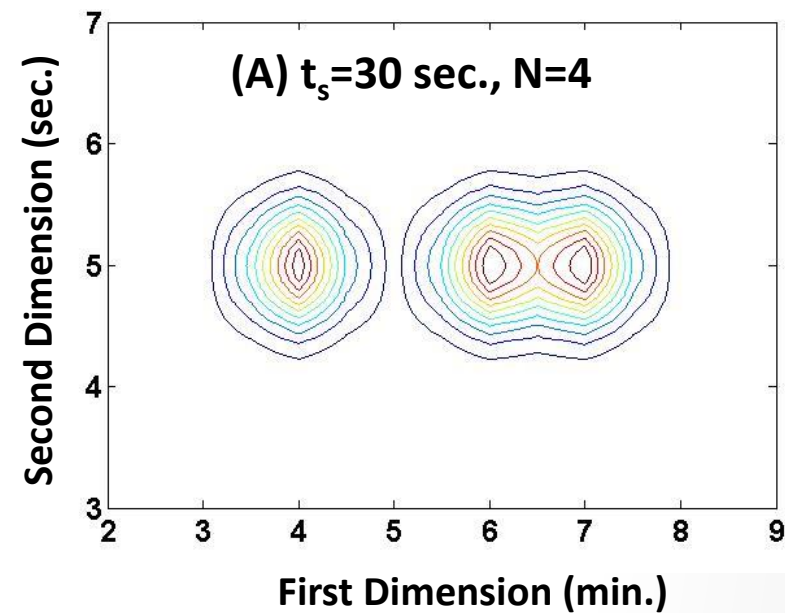
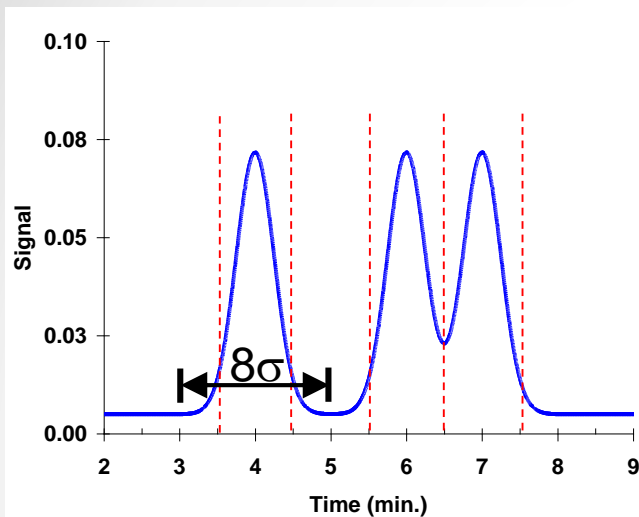
The Murphy-Schure-Foley Criterion



Slide courtesy of Prof. Pete. Carr & Dr. Dwight Stoll

The Undersampling Problem

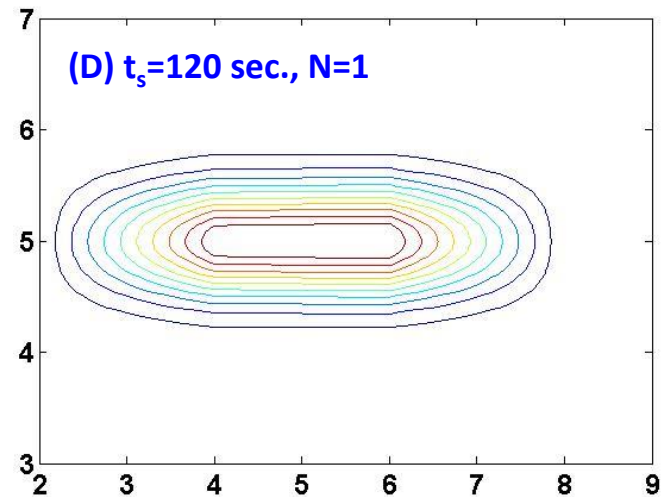
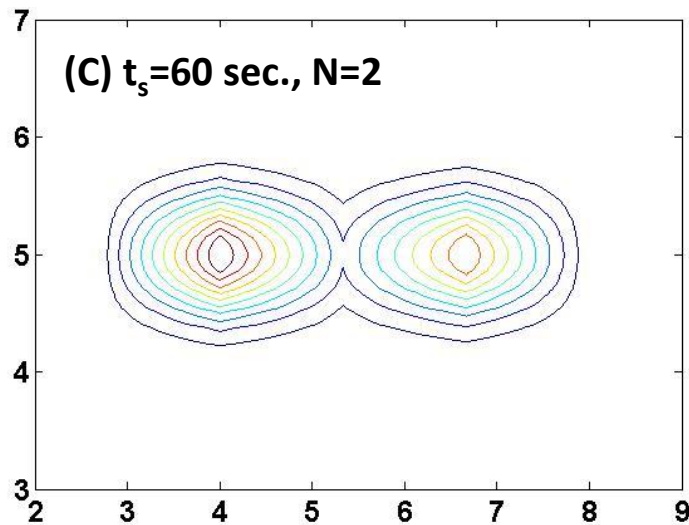
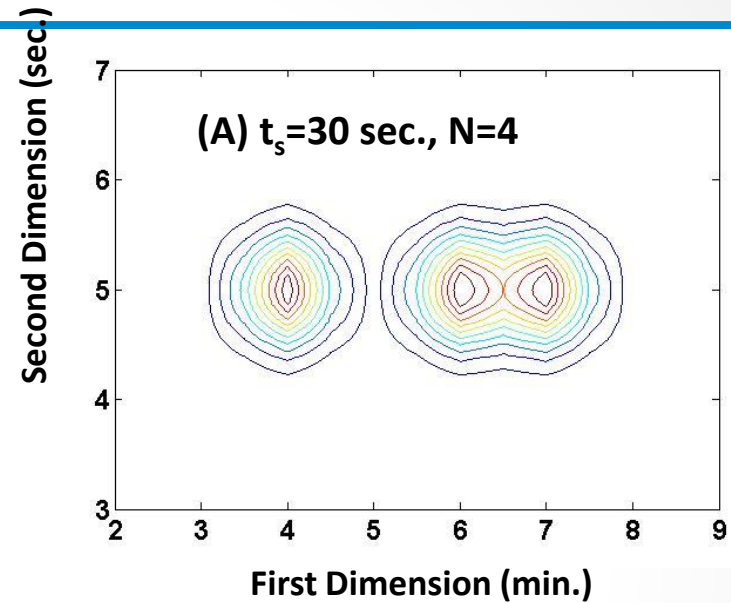
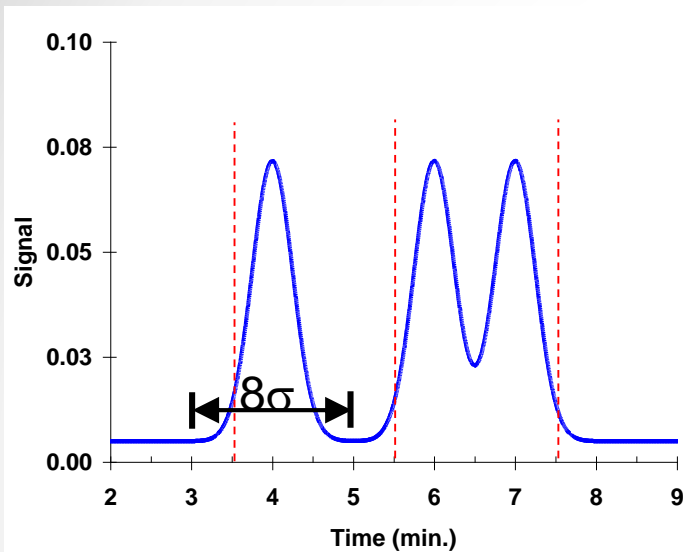
The Murphy-Schure-Foley Criterion



Slide courtesy of Prof. Pete. Carr & Dr. Dwight Stoll

The Undersampling Problem

The Murphy-Schure-Foley Criterion

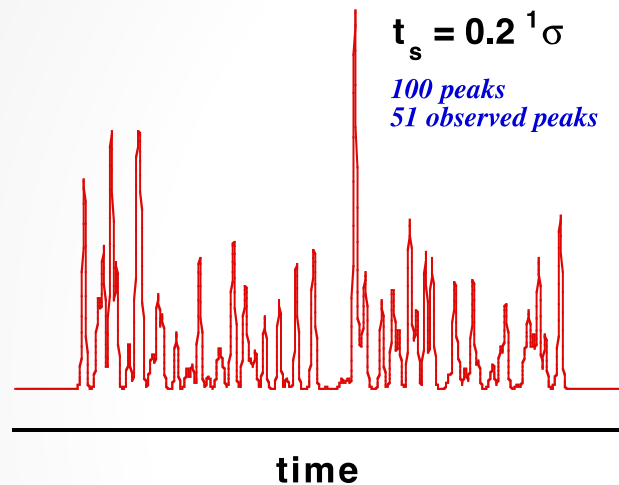


Slide courtesy of Prof. Pete. Carr & Dr. Dwight Stoll

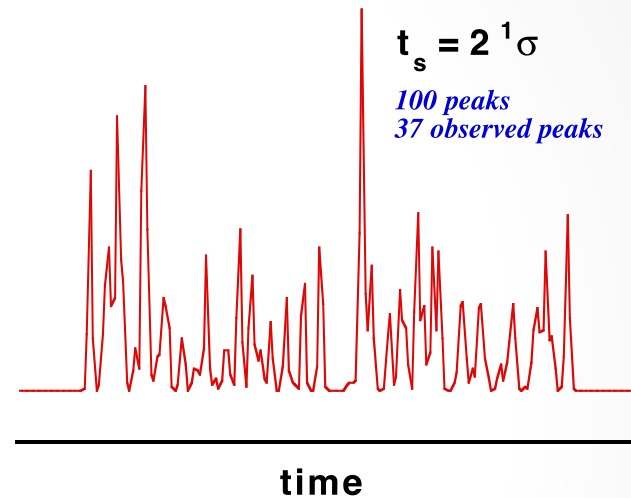
The Undersampling Problem

Alternative View of Undersampling the First Dimension

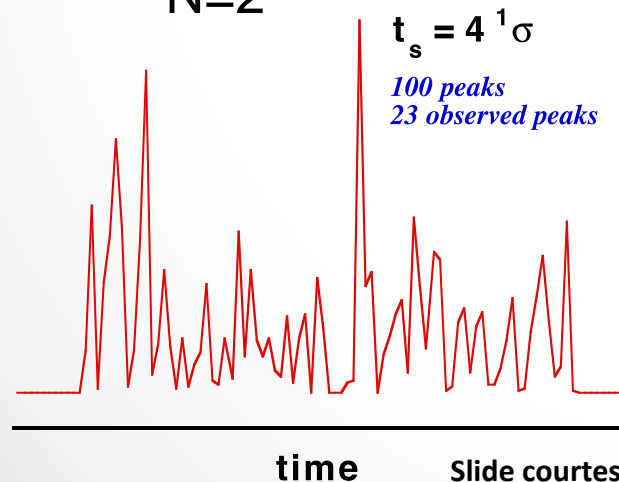
Ideal sampling



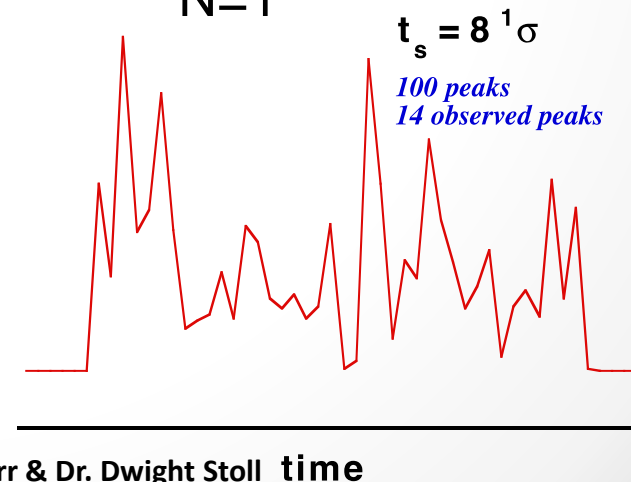
N=4



N=2



N=1



Slide courtesy of Prof. Pete. Carr & Dr. Dwight Stoll

Peak Capacity in Comprehensive 2DLC

"Effective" Peak Capacity

$${}^{2D}Z_p = {}^1Z_p \times {}^2Z_p \times \frac{1}{\langle \beta \rangle} \times f_{coverage} \quad \langle \beta \rangle = \sqrt{1 + 0.214 \cdot \left(\frac{t_s}{\langle {}^1\sigma \rangle} \right)^2}$$

Under Sampling
Correction

Orthogonality

t_s sampling time, ${}^1\sigma$ average
width 1st dimension peaks

1. $\langle \beta \rangle$ average correction for under sampling*
2. $f_{coverage}$ corrects for incomplete use of the separation space.

What is the most important factor?
How can we improve it?

*D.R. Stoll et al., Anal. Chem. 2008, 80, 268-278; Davis, J. M. Stoll, D., R. Carr, P. W. Anal. Chem. 2008, 80(2), 461-473;
 Giddings, J. C. *Multidimensional Chromatography: Techniques and Applications*; Marcel Dekker: New York, 1990
 Slide courtesy of Prof. P. Carr & Dr. D.R.Stoll

Peak Capacity in Comprehensive 2DLC

Example

Assume first dimension is 30 minutes long and has $Z_p = 100$, therefore ${}^1w(4\sigma) = 0.3$ min. If we sample at a rate of $N = 4$ samples/ 8σ we must take a sample every 0.15 minutes (= **9 seconds**) and complete a second dimension chromatogram every 9 seconds.

1. Even so we will lose **27%** of the ideally available peak capacity.
2. Very fast 2D separations are needed in second dimension separation.

$${}^2D Z_p = {}^1Z_p \times {}^2Z_p \times \frac{1}{\langle \beta \rangle} \times f_{coverage}$$

$$\langle \beta \rangle = \sqrt{1 + 0.214 \cdot \left(\frac{t_s}{\langle {}^1\sigma \rangle} \right)^2}$$

Further reading:

K. Horie et al., *Analytical Chemistry*. **2007**, 79, 3764–3770.

J Seeley, J. *Journal of Chromatography A*. **2002**, 962, 21–27.

L.M. Blumberg, *Journal of Separation Science*. **2008**, 31, 3358–3365.

L.M. Blumberg, et al., *P. Journal of Chromatography A*. **2008**, 1188, 2–16.

Peak Capacity in Comprehensive 2DLC

Implications of β

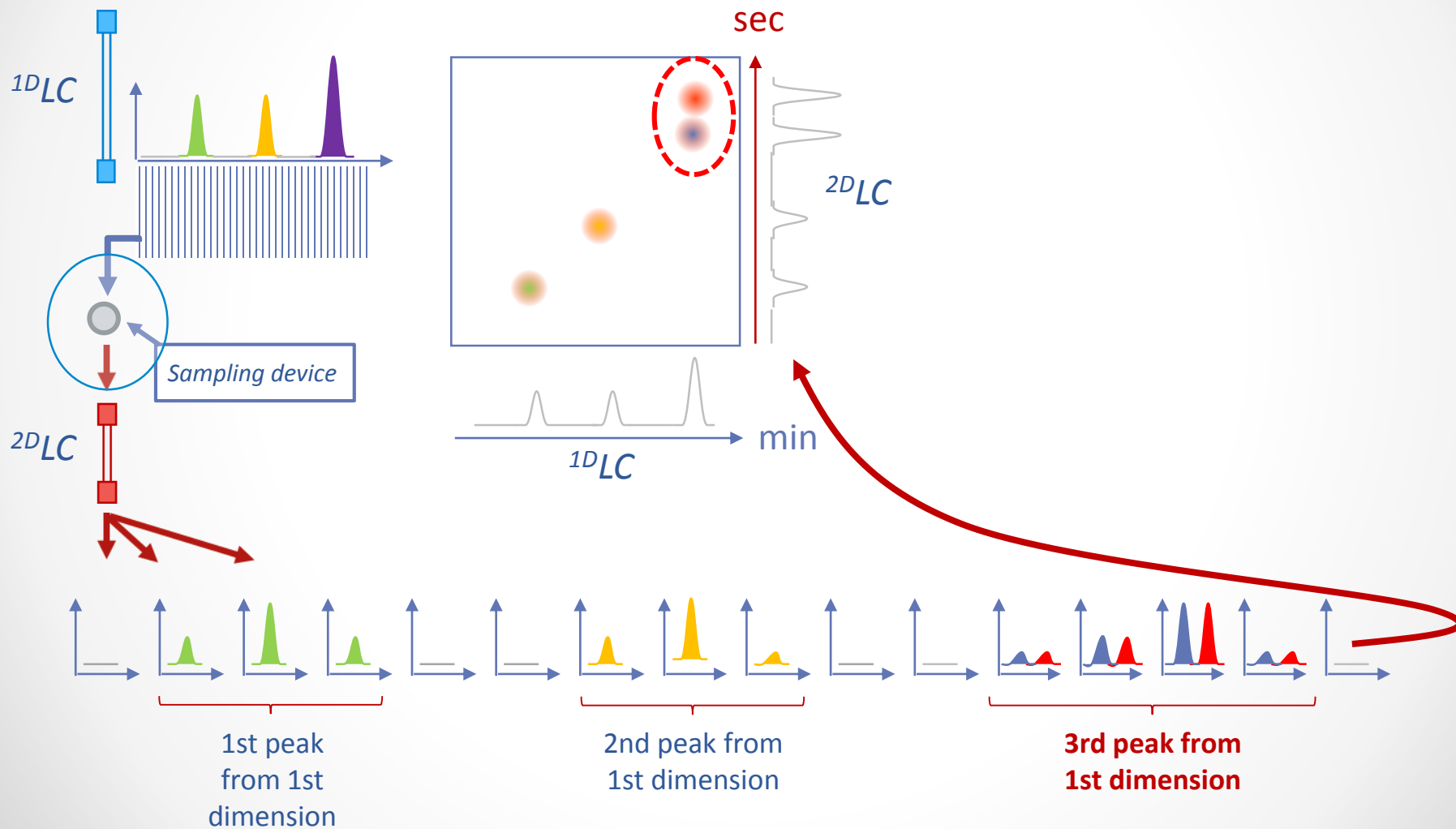
- We want to make the sampling time short.
- In LC x LC ${}^1t_{sample} = {}^2t_{cycle}$
- Prefer ${}^1t_{sample} < {}^2t_{cycle}$ (under fill the sample loop!)
- ${}^2t_{cycle} = {}^2t_{gradient} + {}^2t_{re-equilibration}$
- Don't make ${}^1t_{sample}$ too short since 2D separation peak capacity decreases if ${}^2t_{gradient}$ decreases
- Clearly there is an optimum range in $t_{sample} ({}^2t_{cycle})$

Practical Implementation for 2D HPLC

Sampling Device, Column Selection

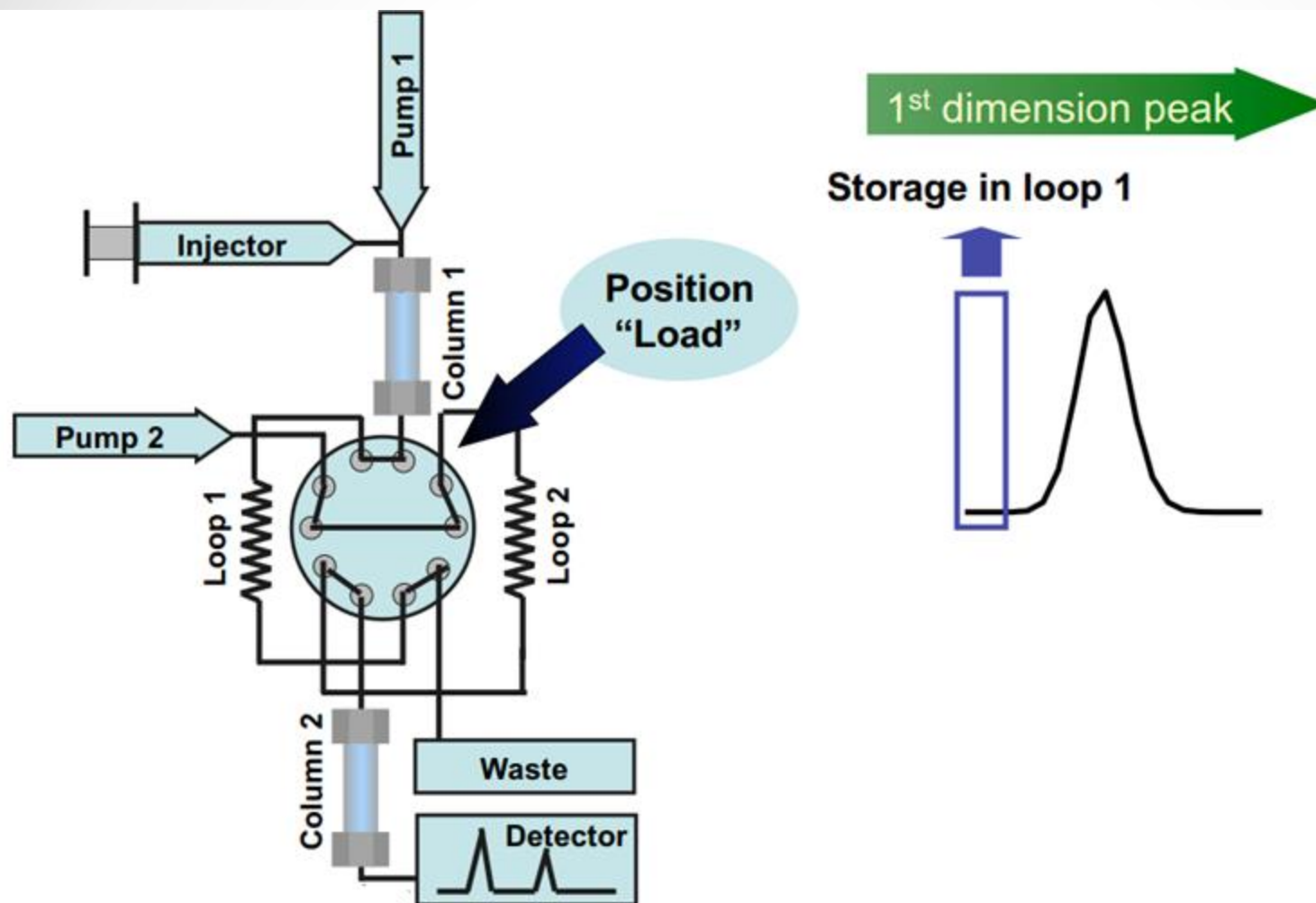
Sampling Device for LCxLC

First-In First-Out (FIFO) Configuration



Sampling Device for LCxLC

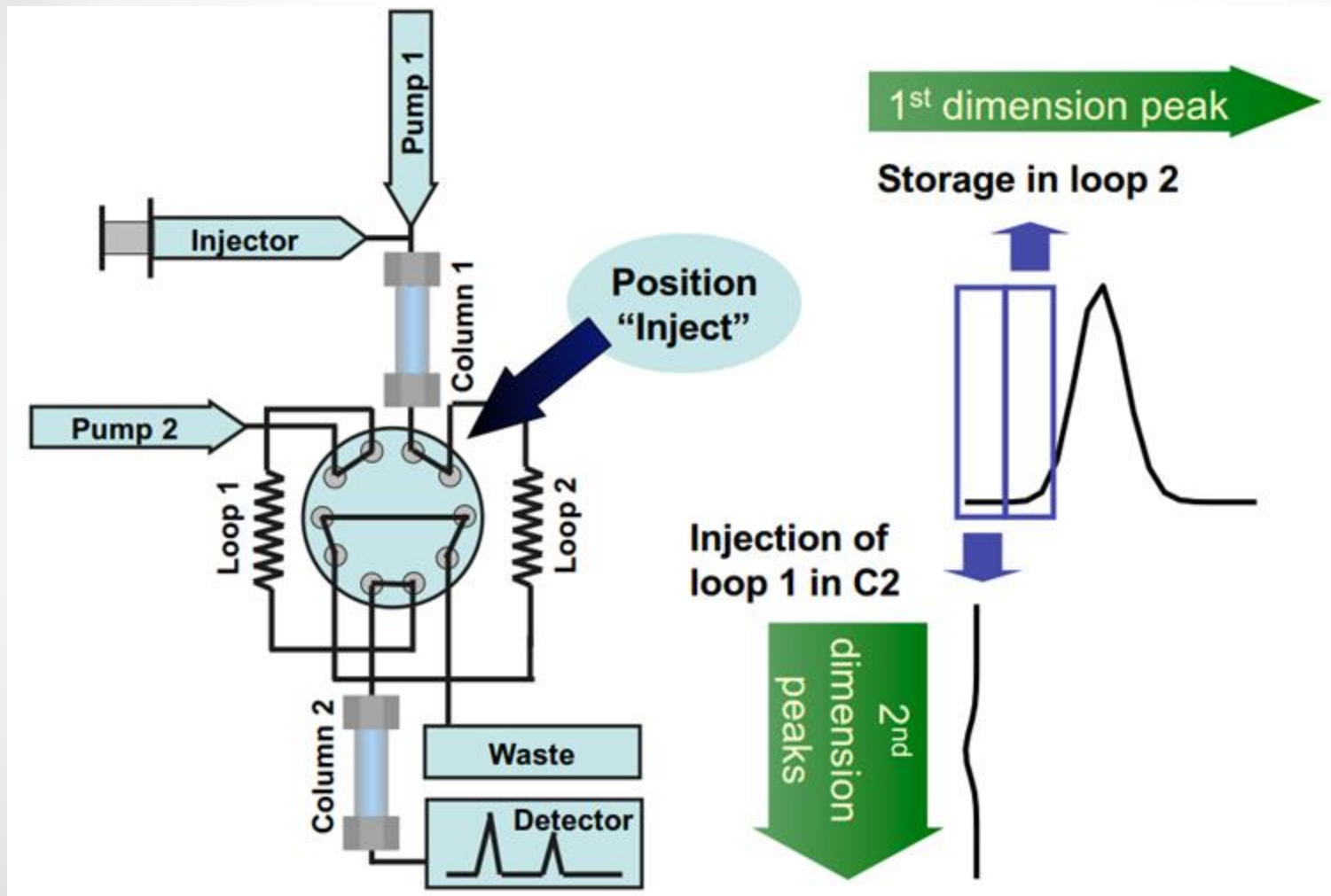
First-In First-Out (FIFO) Configuration (10 port, 2 position valve)



Slide courtesy of Prof. P. Schoenmakers

Sampling Device for LCxLC

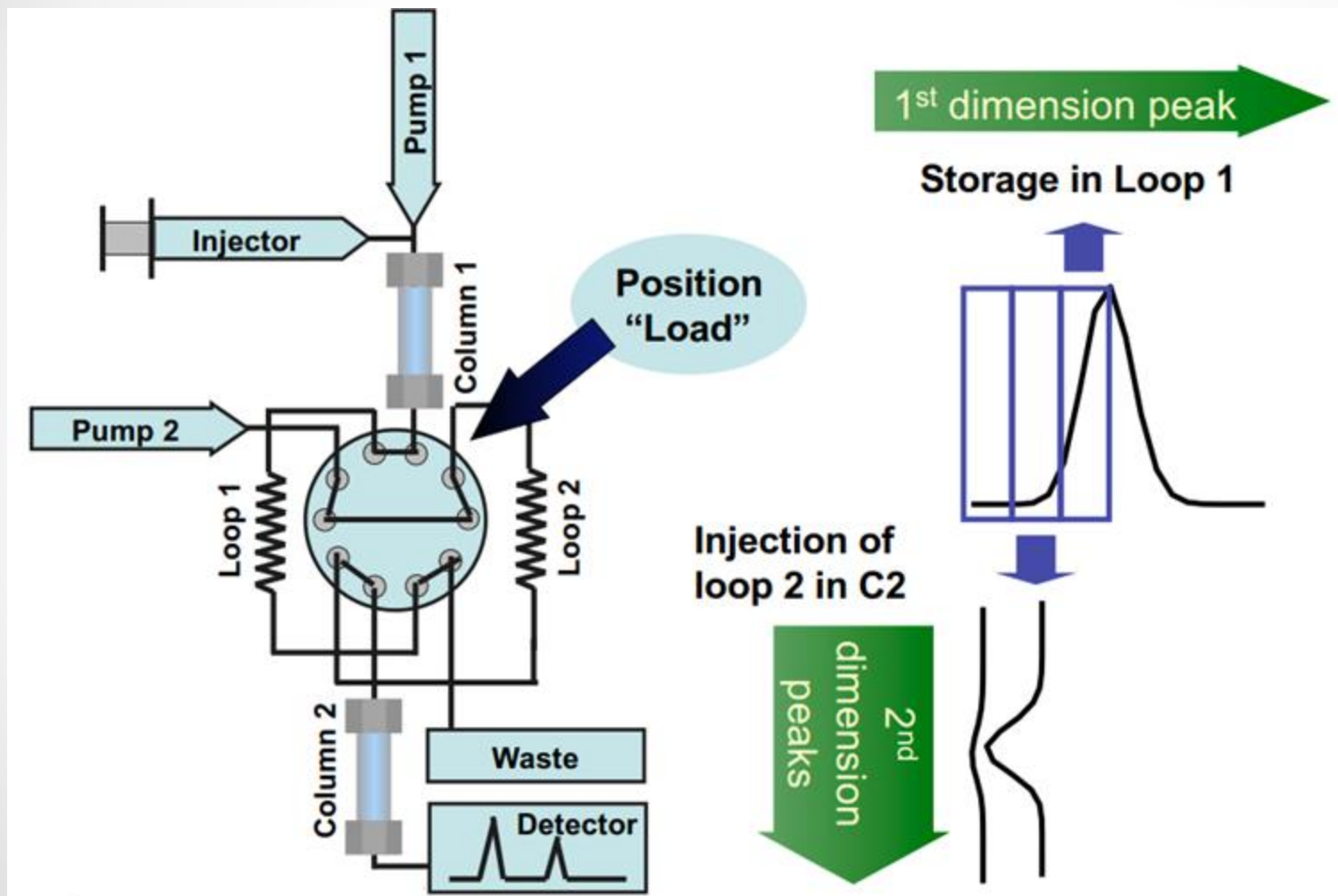
First-In First-Out (FIFO) Configuration(10 port, 2 position valve)



Slide courtesy of Prof. P. Schoenmakers

Sampling Device for LCxLC

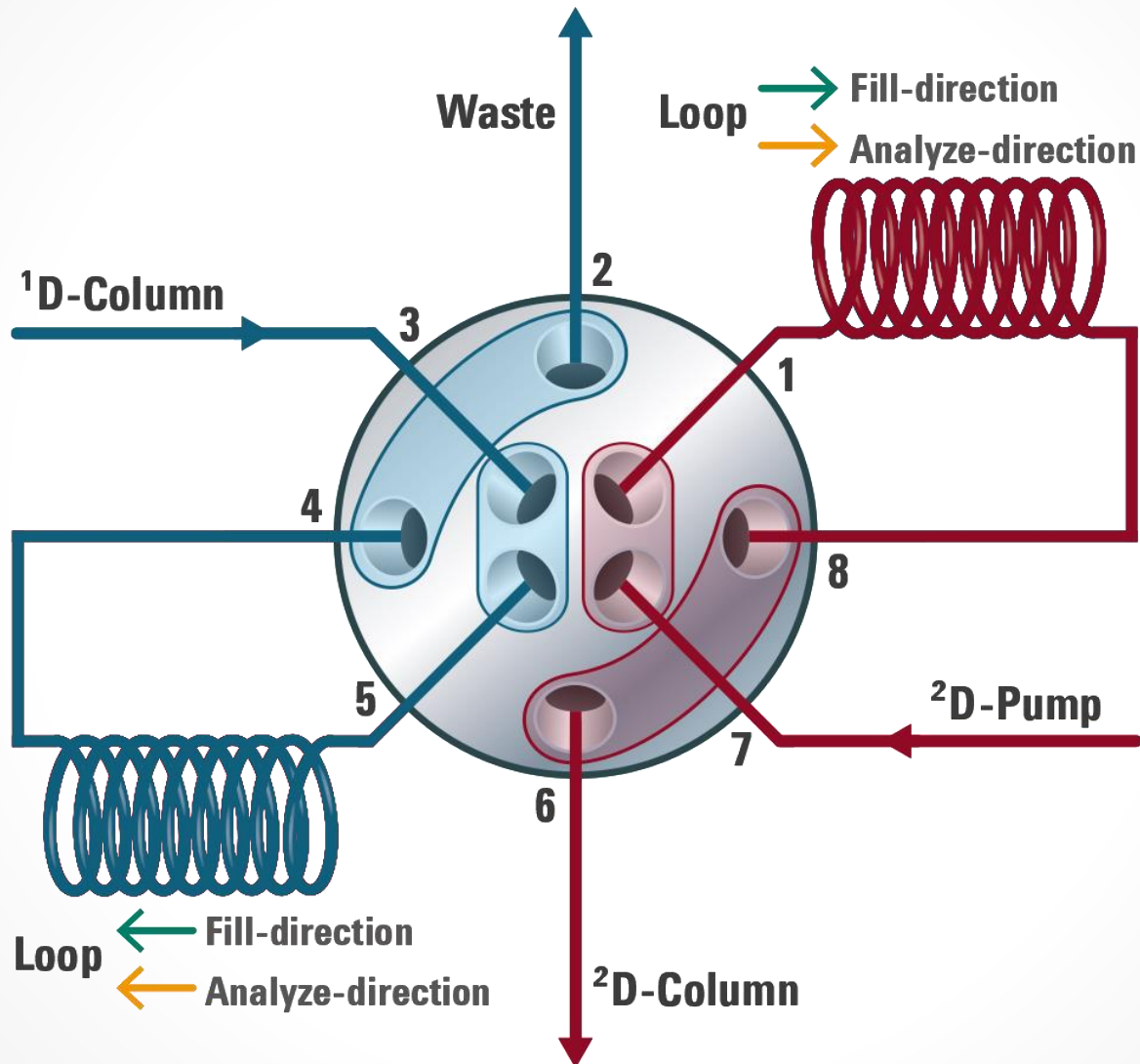
First-In First-Out (FIFO) Configuration (10 port, 2 position valve)



Slide courtesy of Prof. P. Schoenmakers

Sampling Device for LCxLC

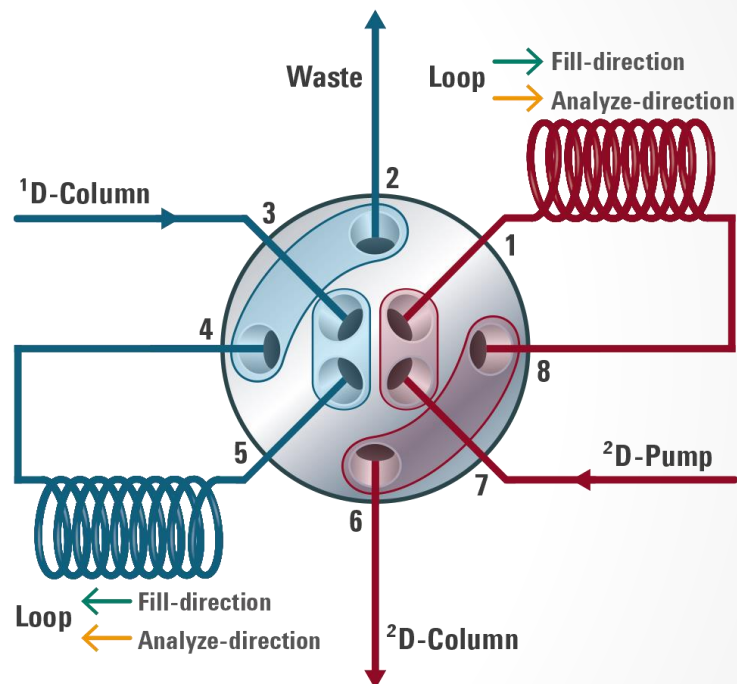
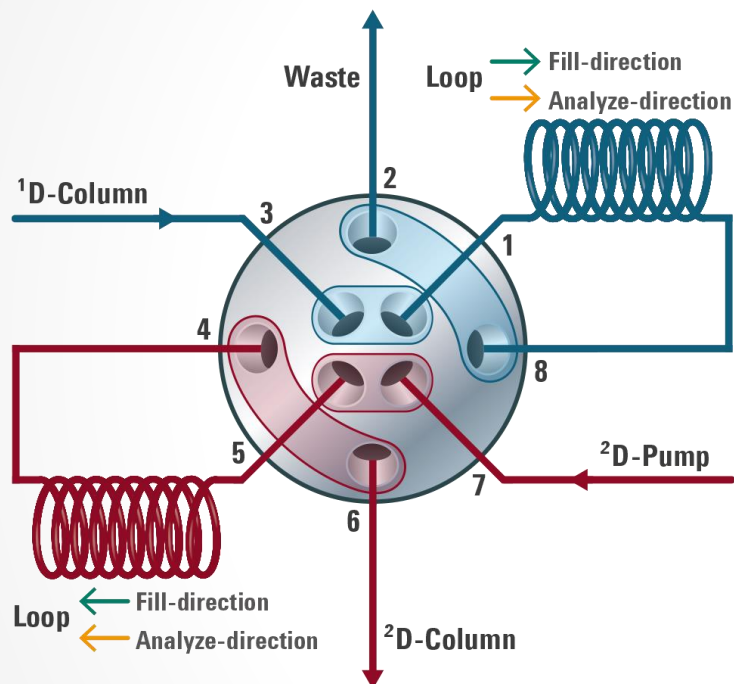
2x4 port, 2 position valve (Agilent Technologies Duo Valve)



Slide courtesy of Agilent Technologies

Sampling Device for LCxLC

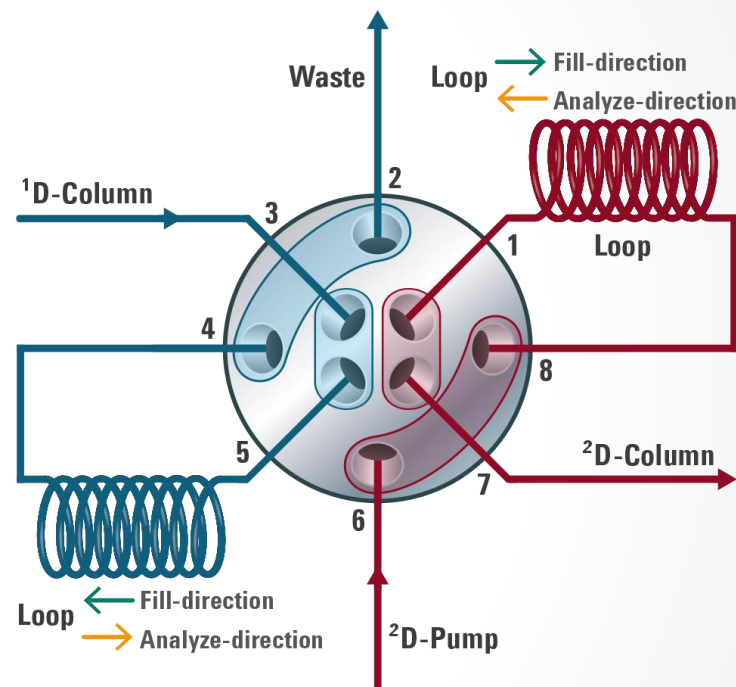
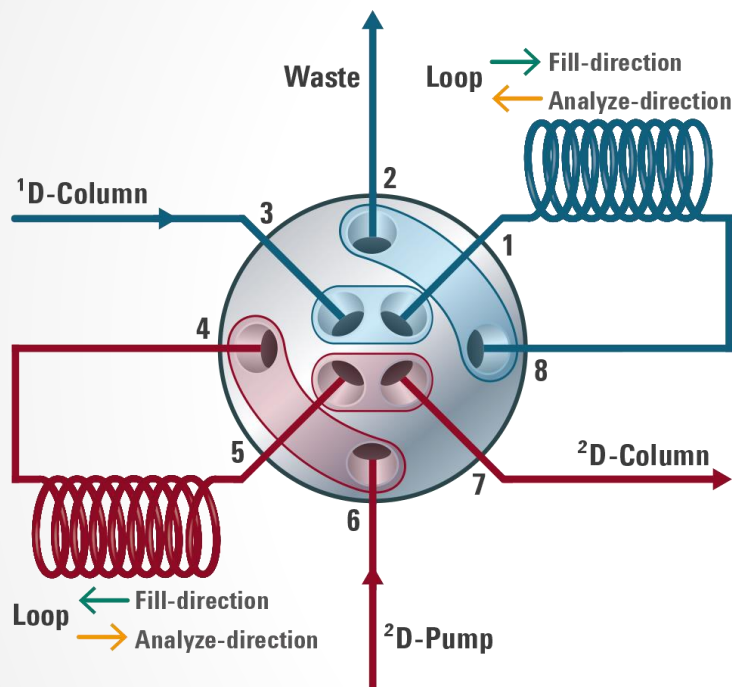
2x 4 port, 2 position valve, co-current mode (Agilent Technologies Duo Valve)



First-In-First-Out (FIFO)

Sampling Device for LCxLC

2x 4 port, 2 position valve, counter-current mode (Agilent Technologies Duo Valve)

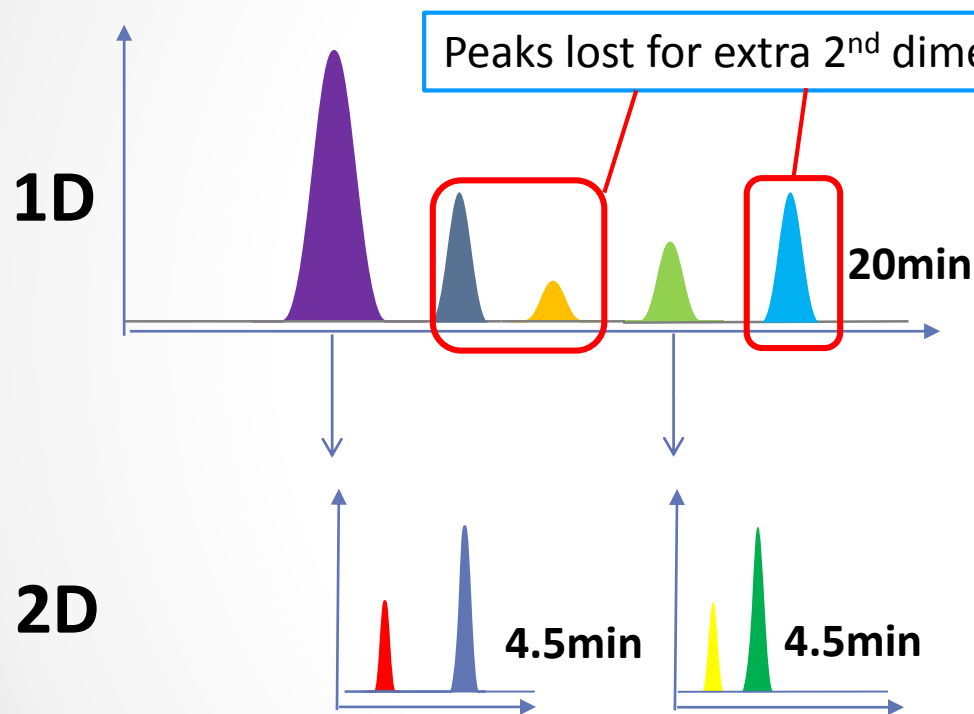


First-In-Last-Out (FIFO)

Counter-Current Mode: connections on port 6 and 7 reversed!

Sampling Device for LC-LC (Heart-Cut)

Long Analysis Time of 2nd Dimension Separation



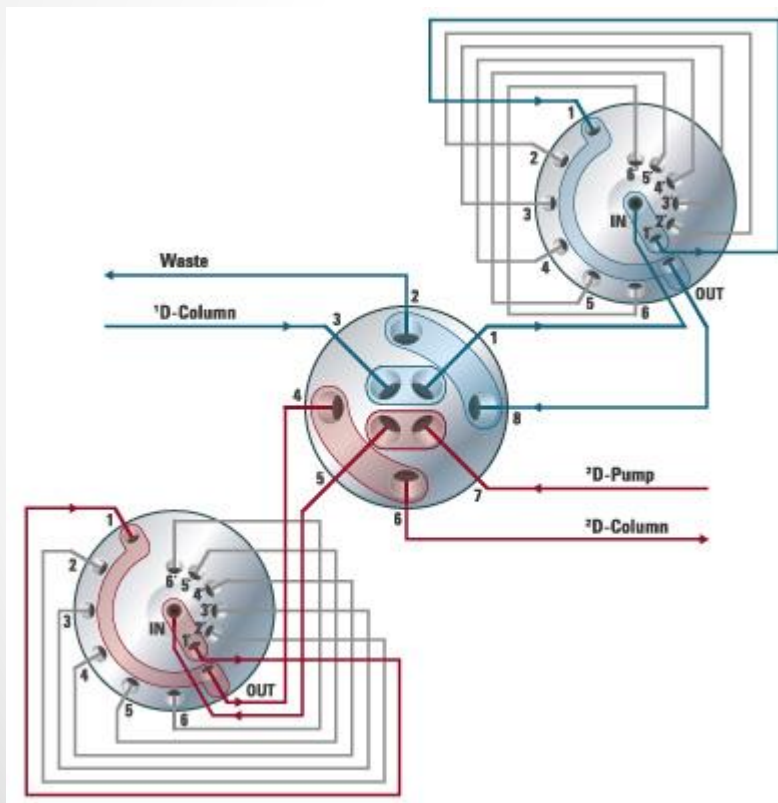
Heart-cutting Data Viewer

Slide courtesy of Agilent Technologies

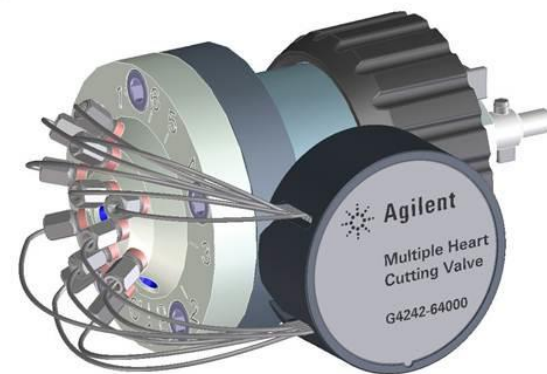
Sampling Device for LC-LC (Heart-Cut)

Agilent Multiple Heart-Cutting 2D-LC

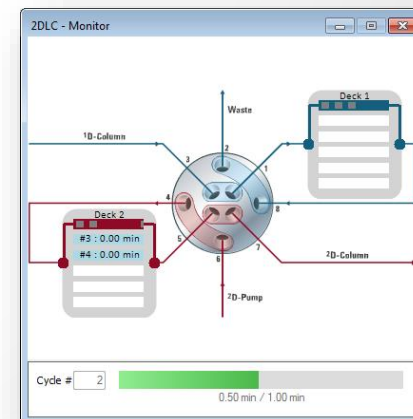
Smart Valve-Loop Setup with 12 loops
→ 2D-LC valve + two 6/14 valves



Pre-aligned loop-valve kits, just add to the existing 2D-LC system



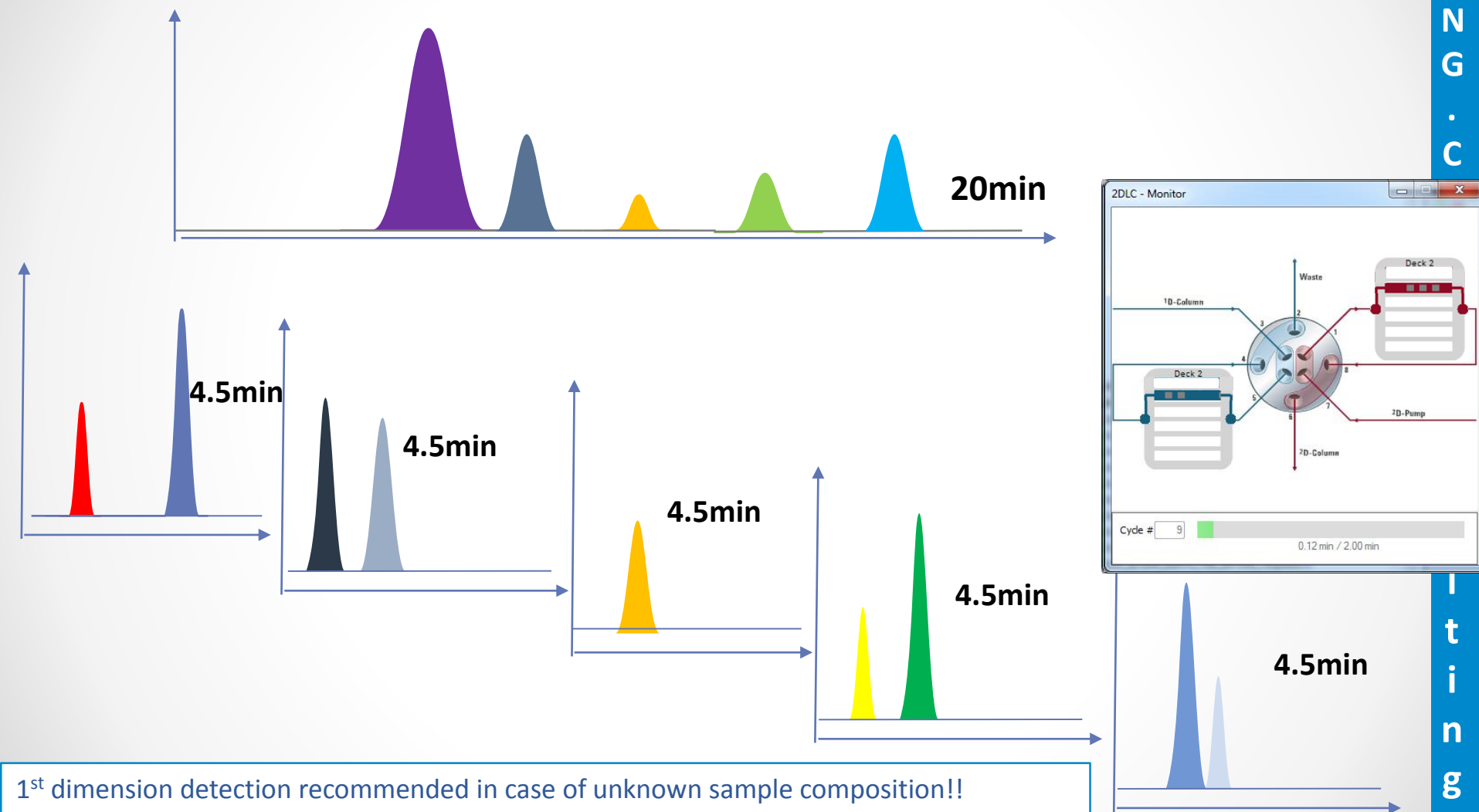
Online status monitoring



Slide courtesy of Agilent Technologies

Sampling Device for LC-LC (Heart-Cut)

Agilent Multiple Heart-Cutting 2D-LC



Slide courtesy of Agilent Technologies

End of Part 1